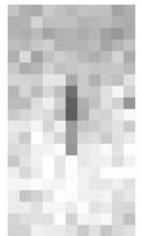
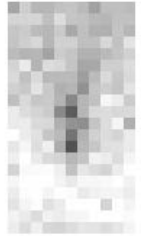
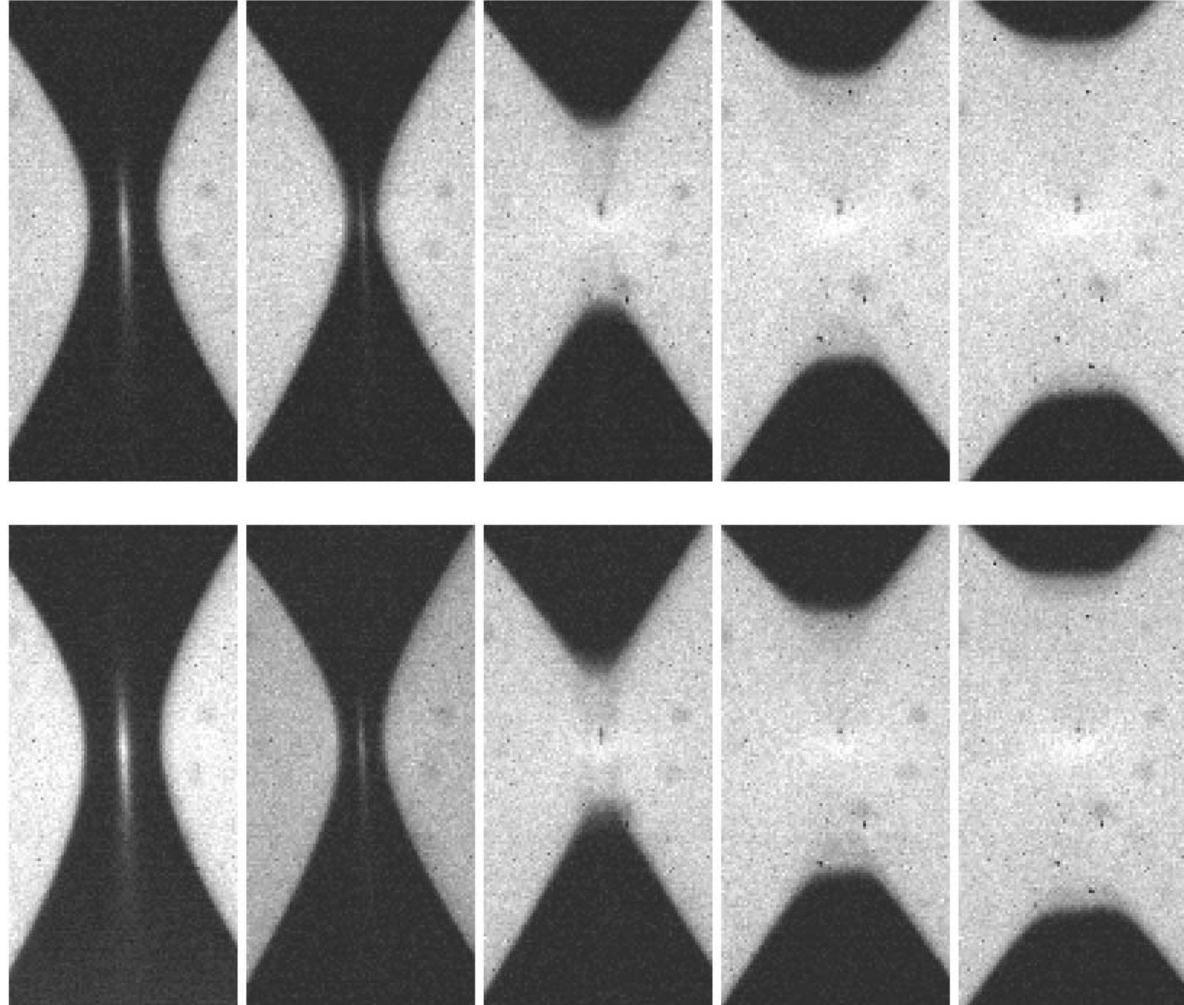


8 times viscosity of water

50 μm



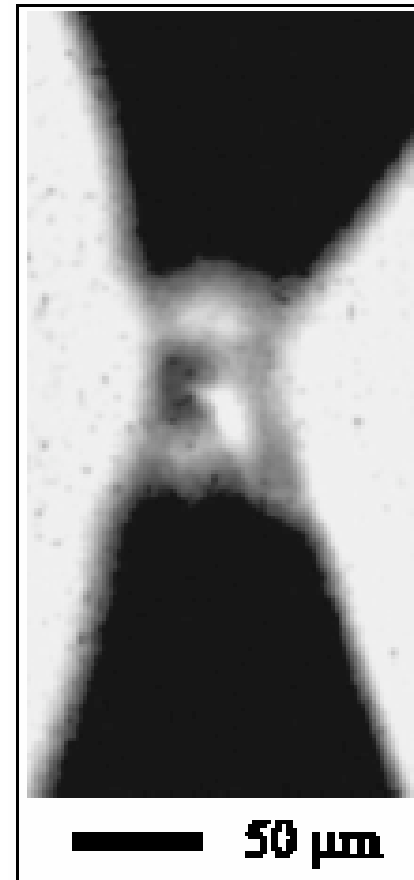
$dt = 5 \mu\text{s}$



Neck Rupture?

Burton, Waldrep & Taborek
Phys. Rev. Lett., **94**,
Paper no. 184502 (2005)

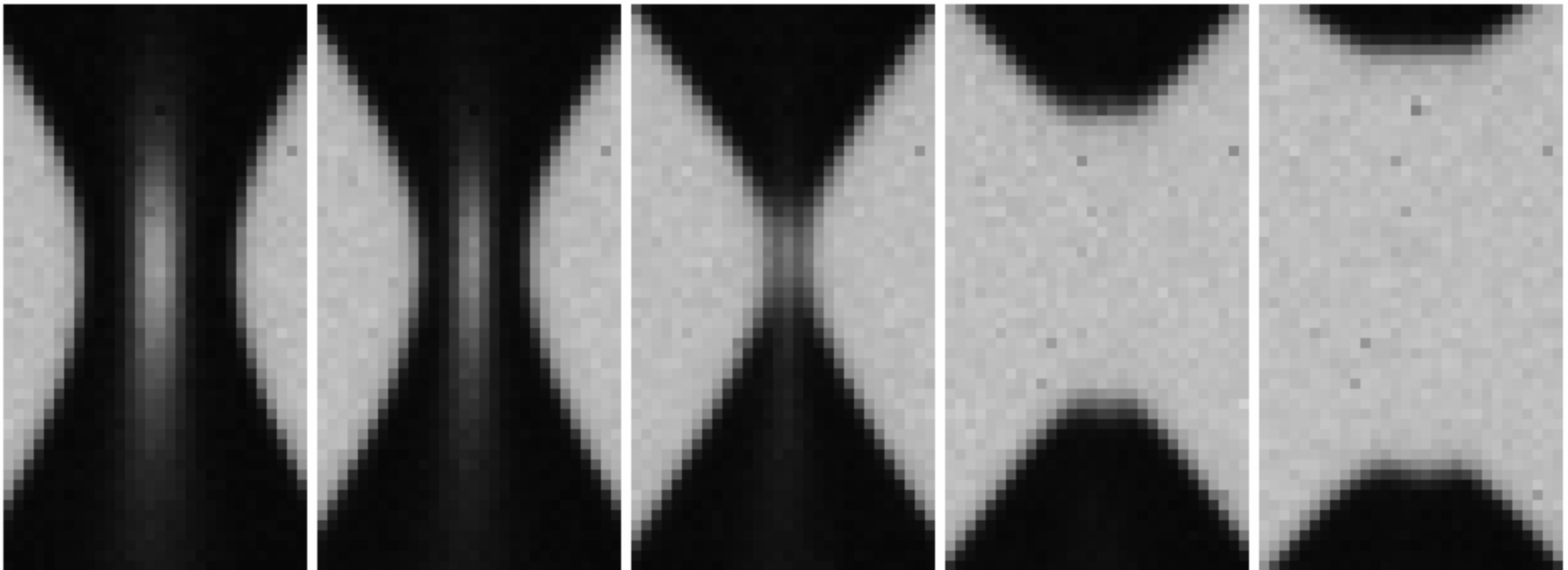
100,000 fps



Helium

50 μm

$dt = 10 \mu\text{s}$

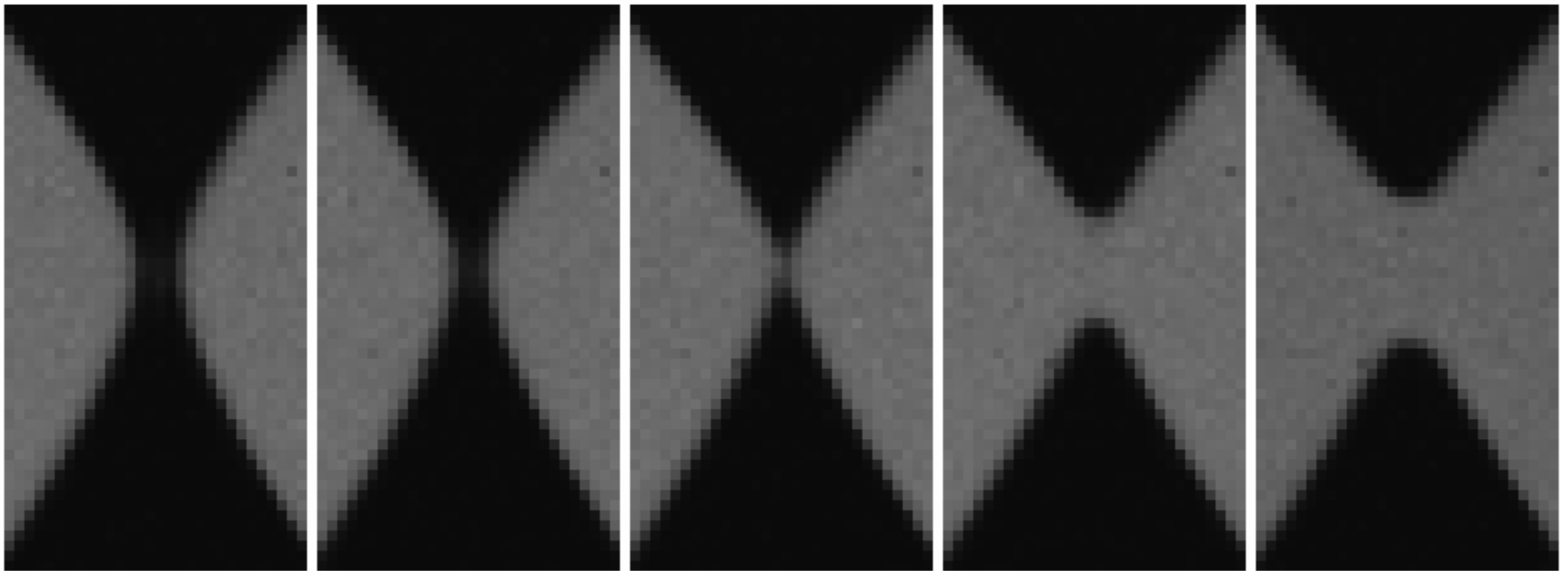


100 000 fps

Helium in water

50 μm

$dt = 1 \mu\text{s}$

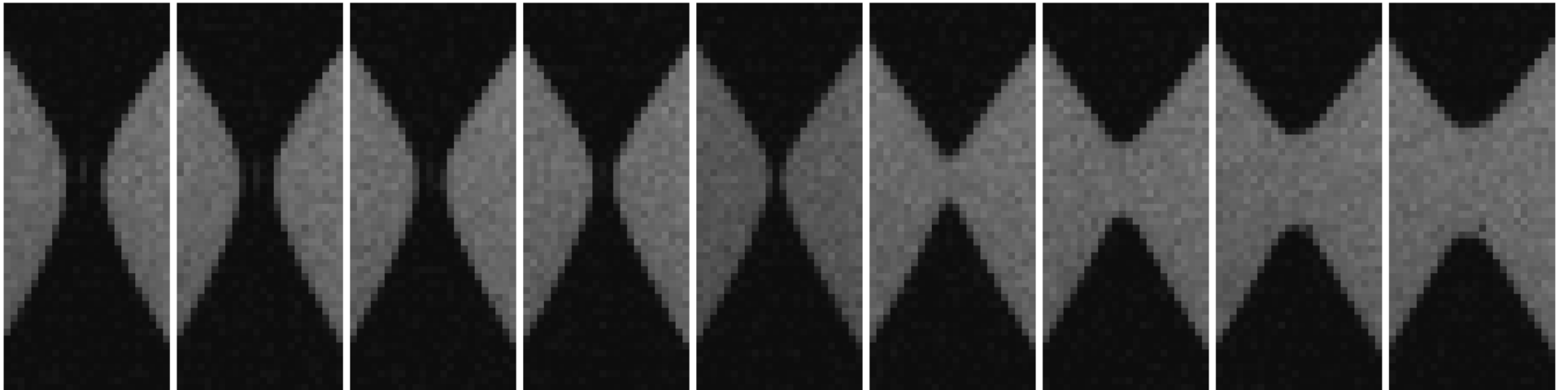


1 000 000 fps

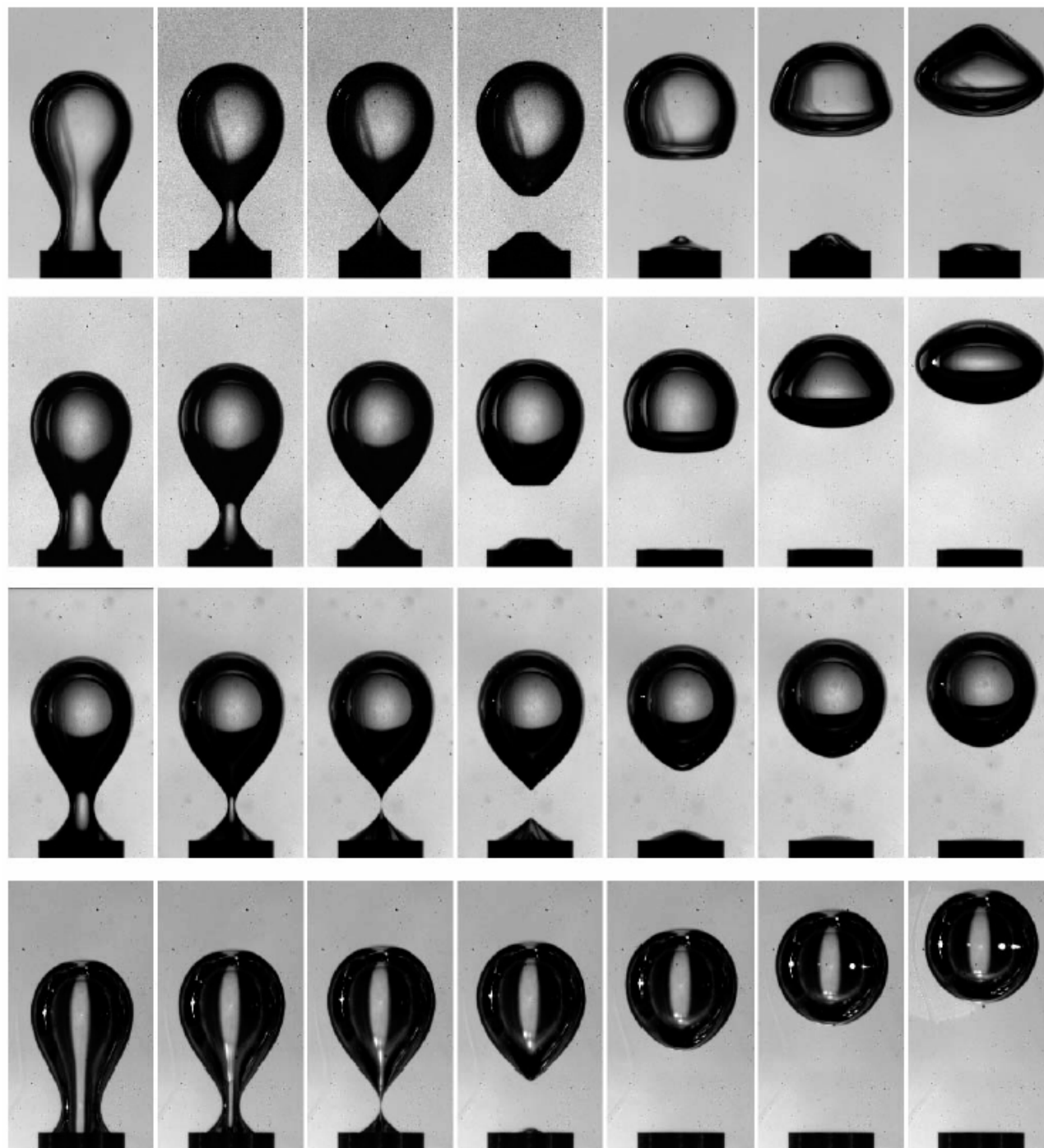
Air in water

50 μm

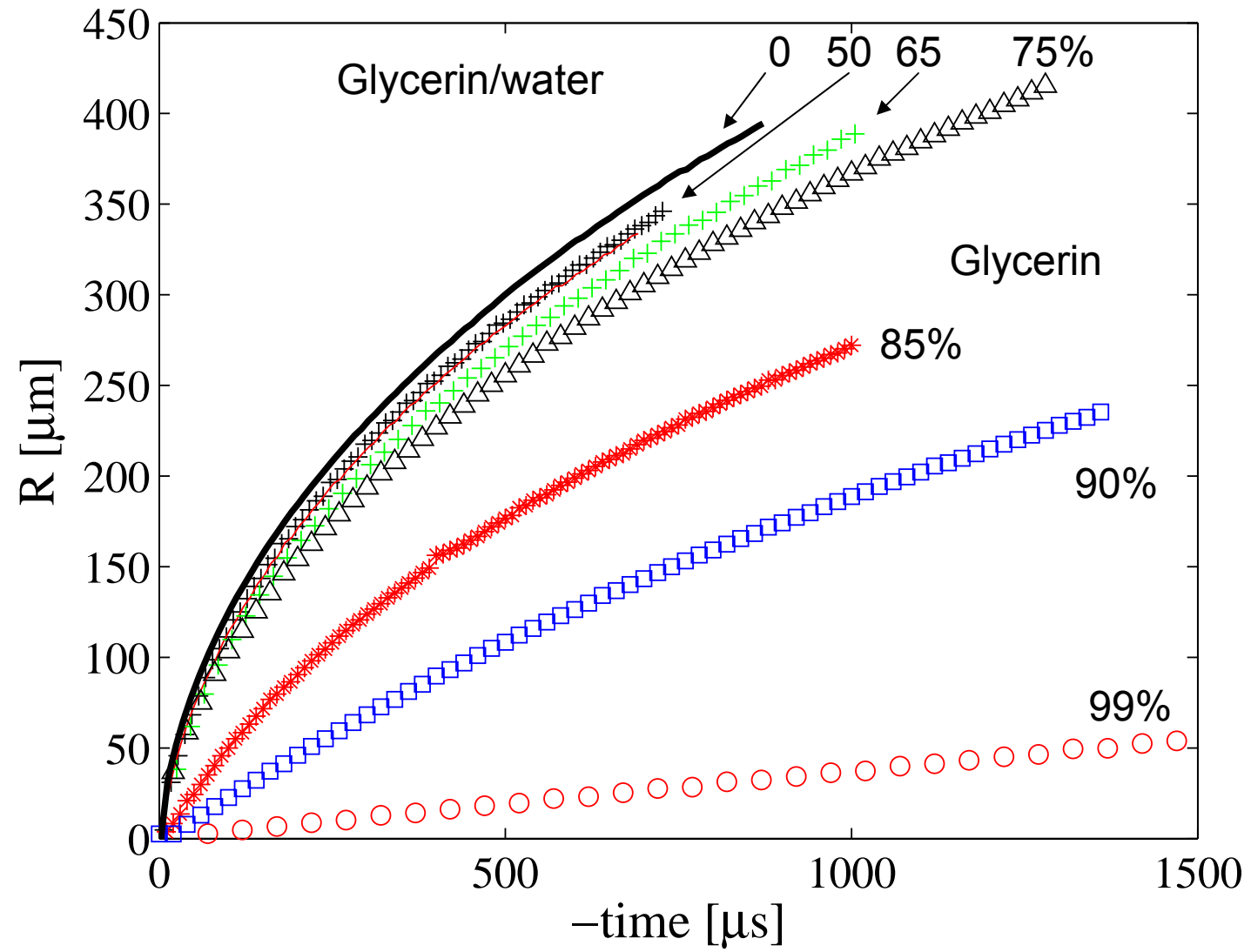
$dt = 1 \mu\text{s}$

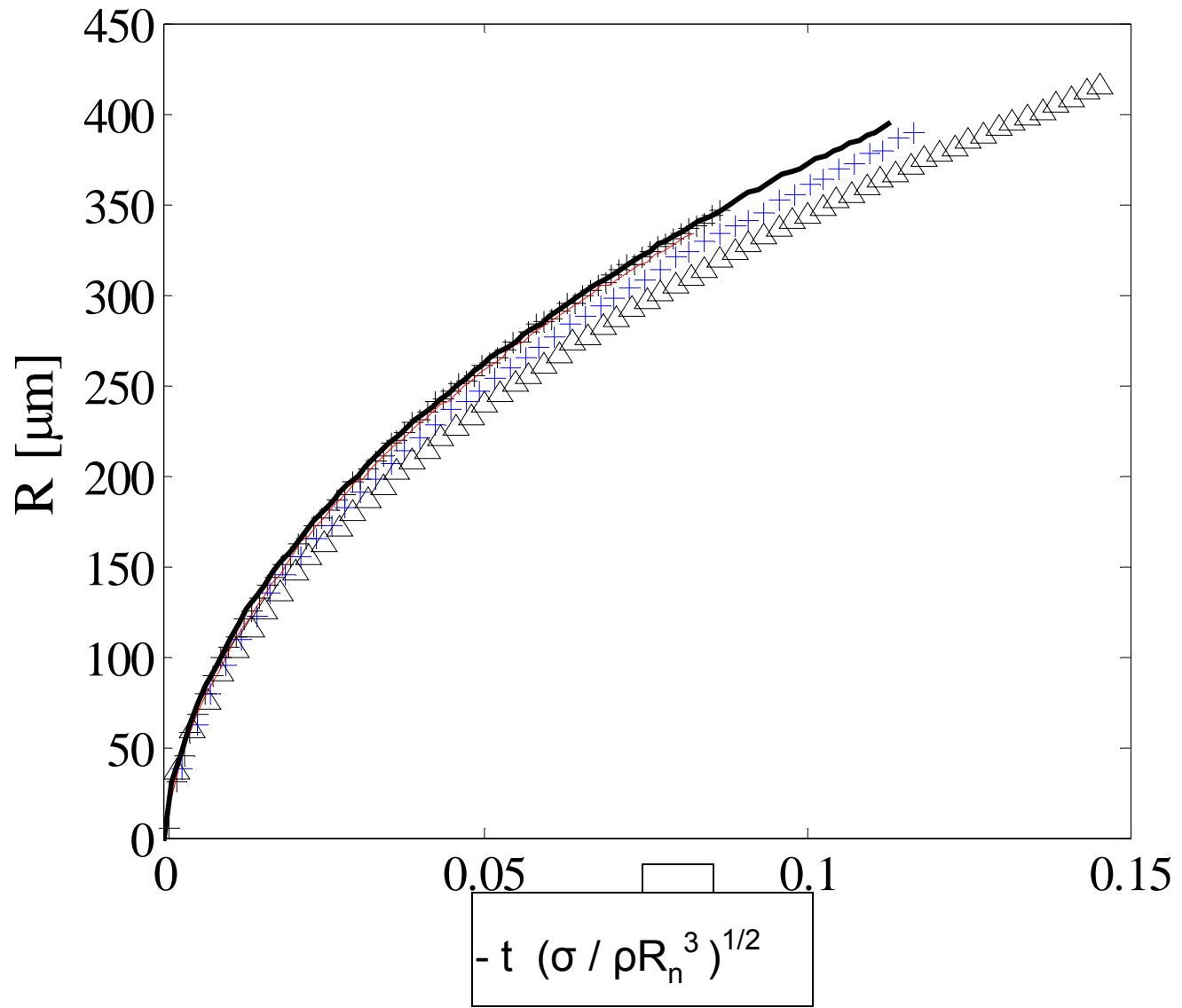


Increasing
viscosity



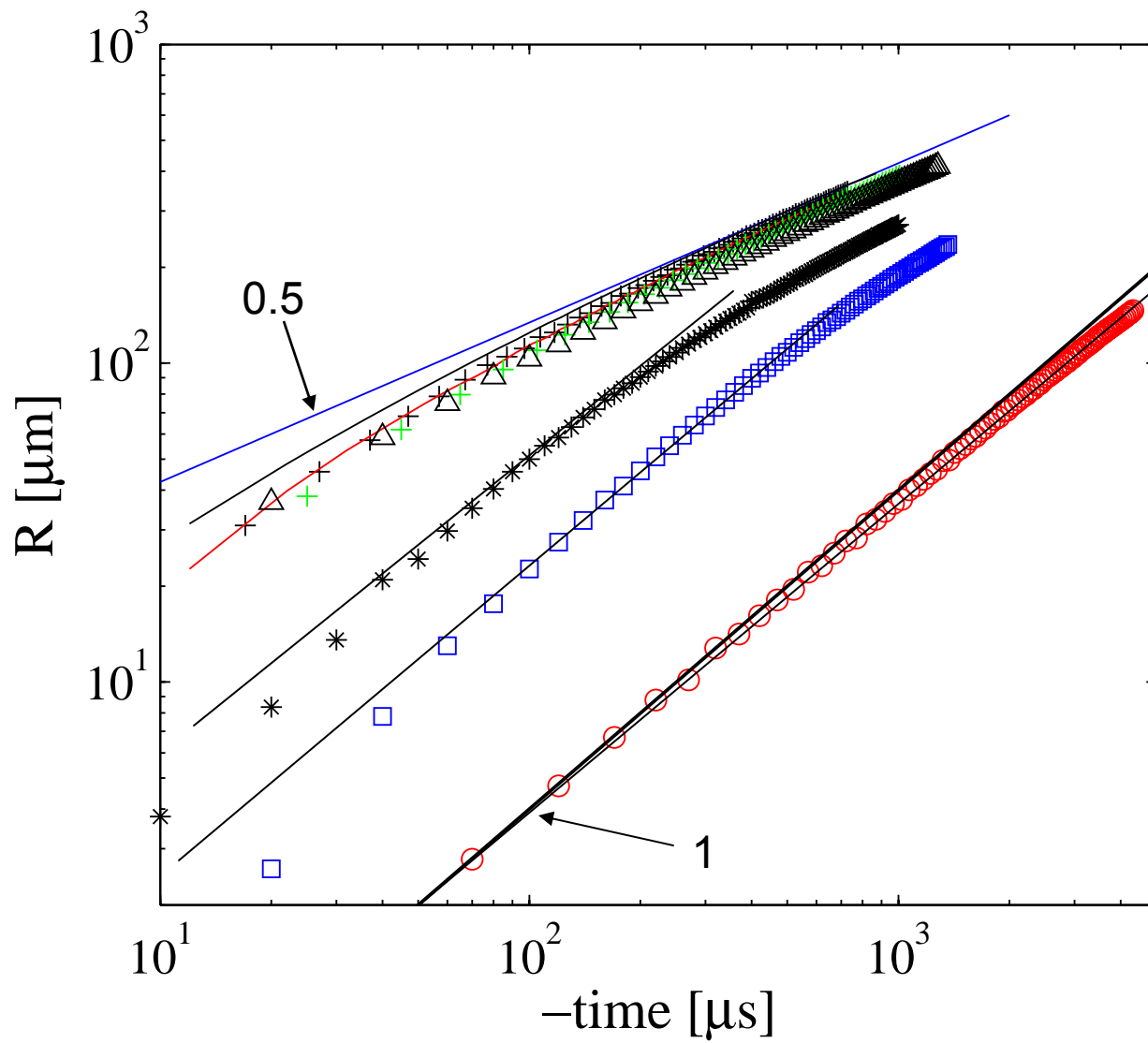
Higher liquid viscosity



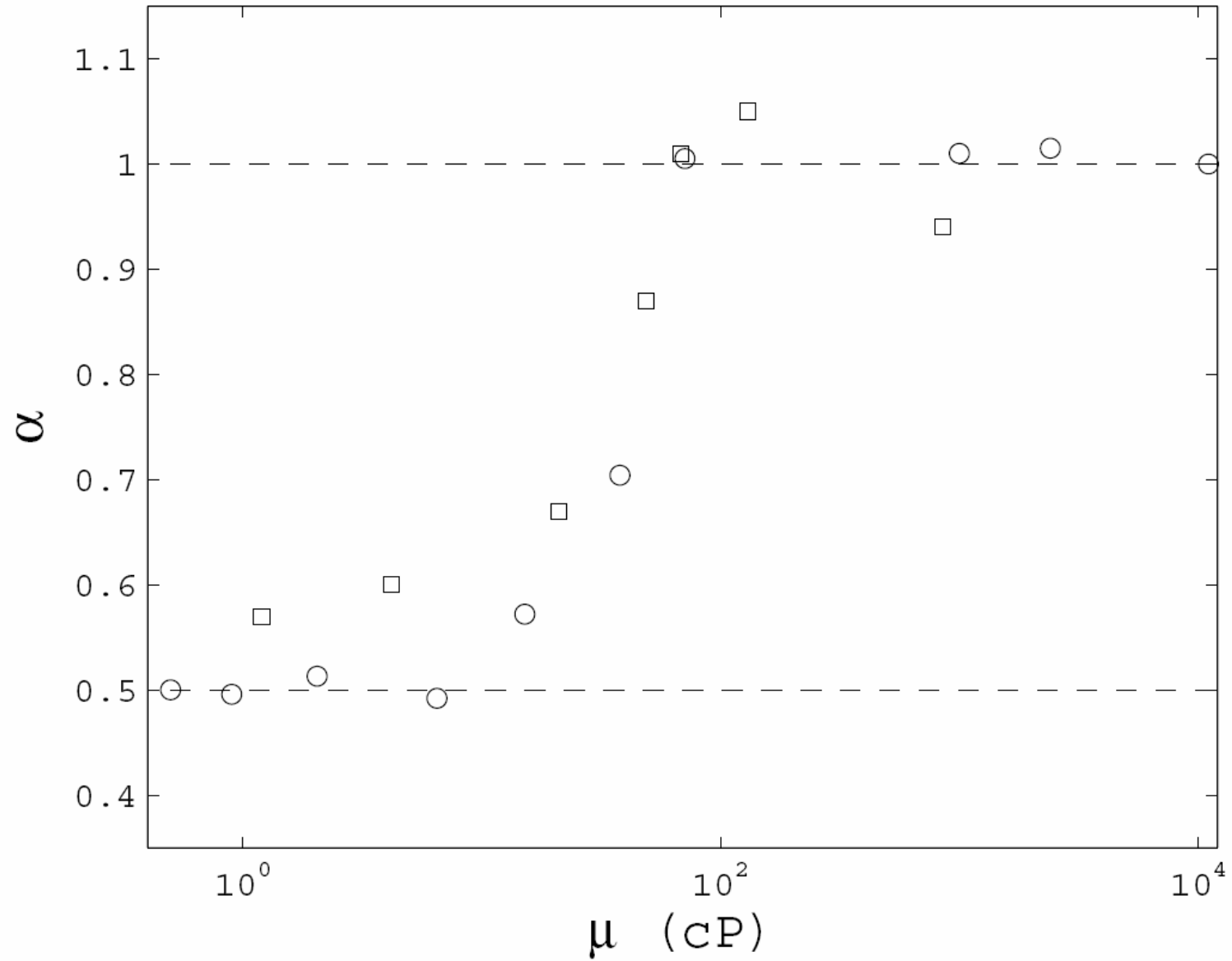


Deviates at
 $\mu \approx 10$ water

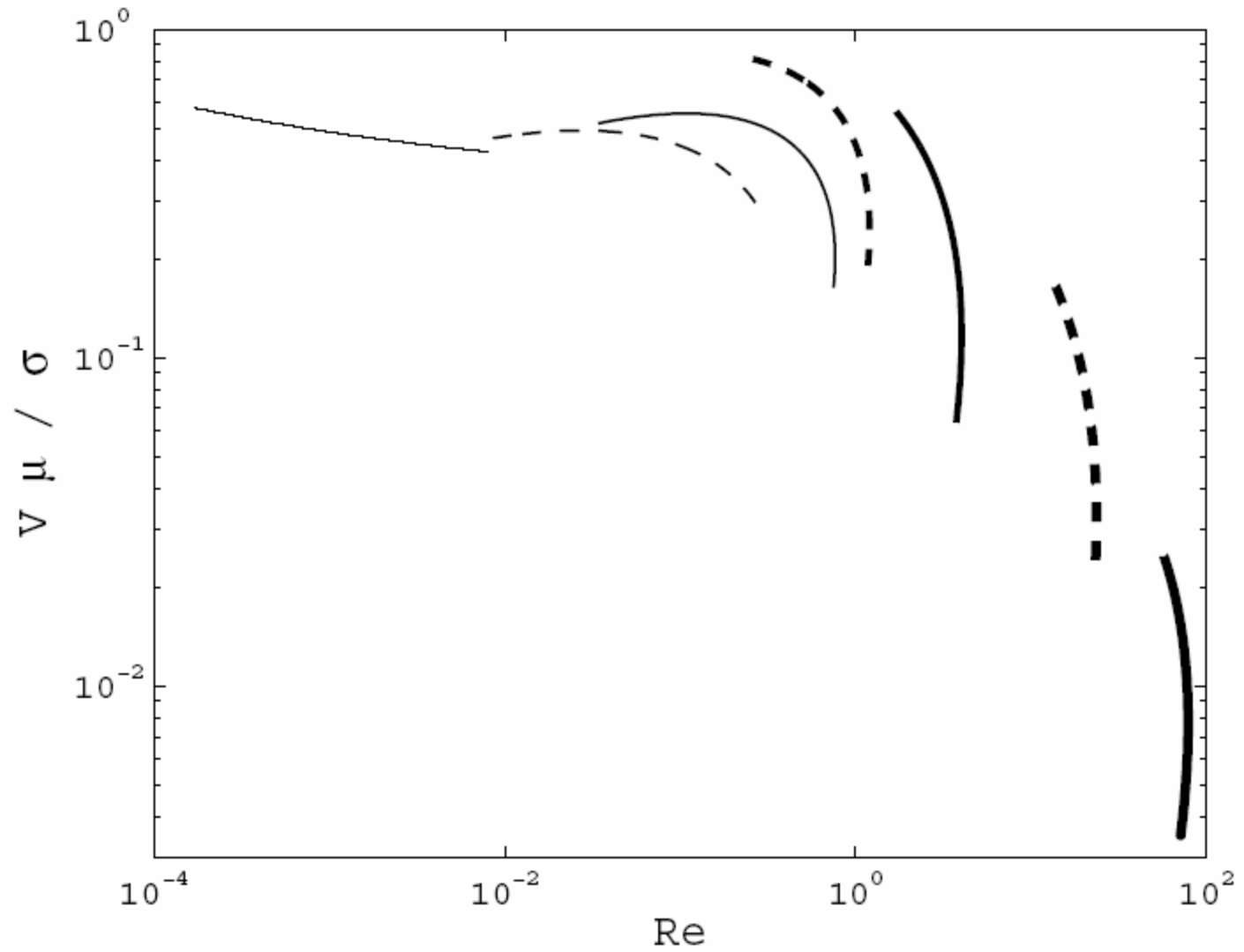
Power-laws ?

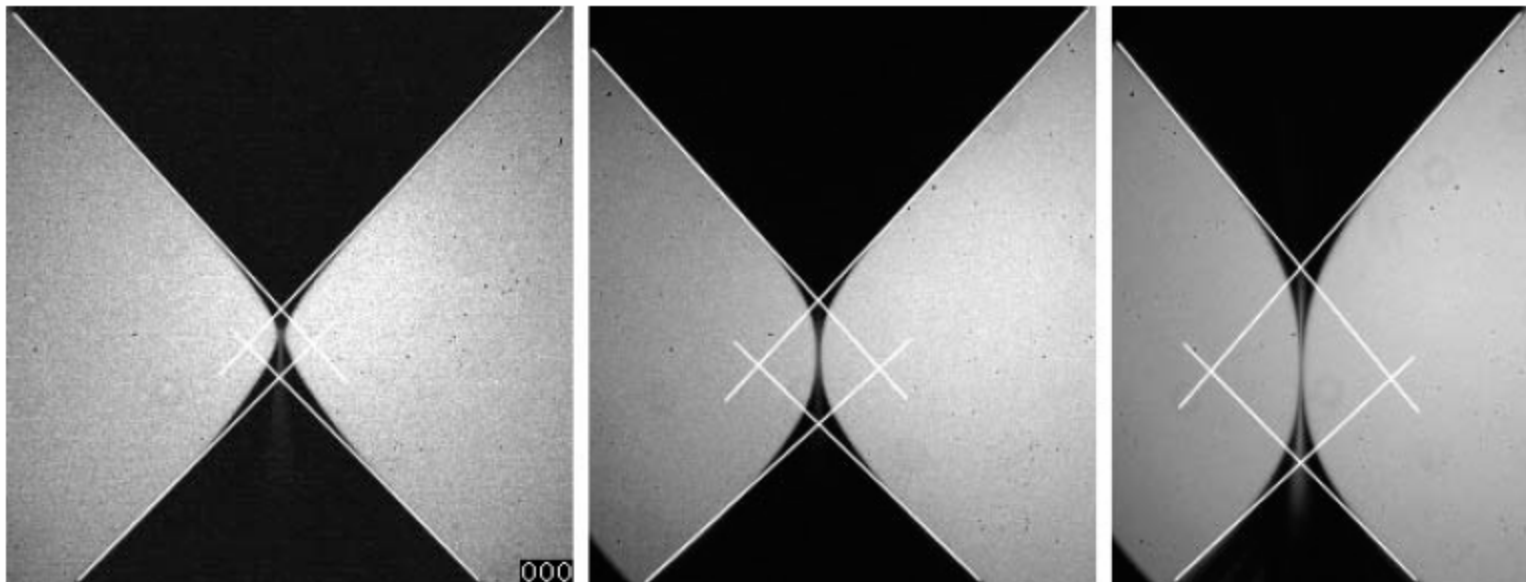


Power-law exponent, Compared to Burton et al. (2005)

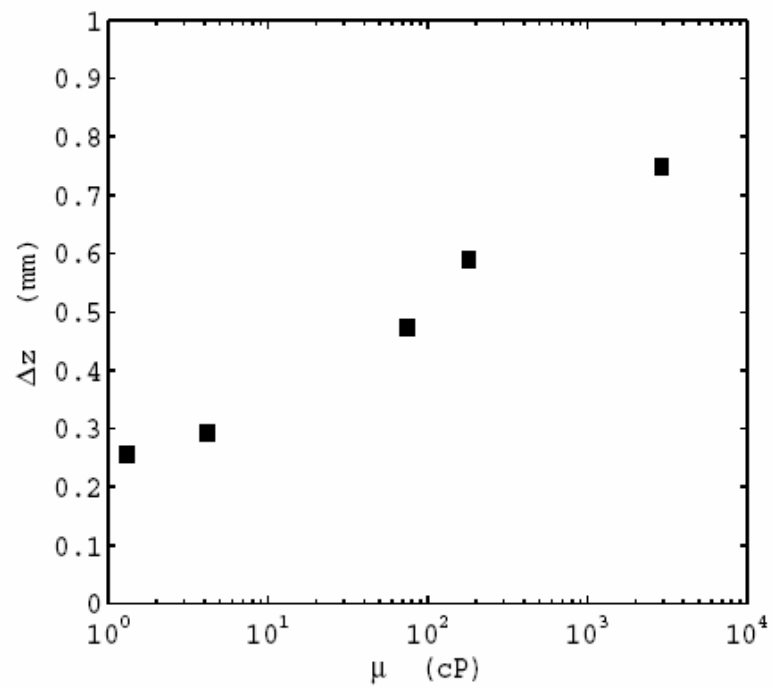
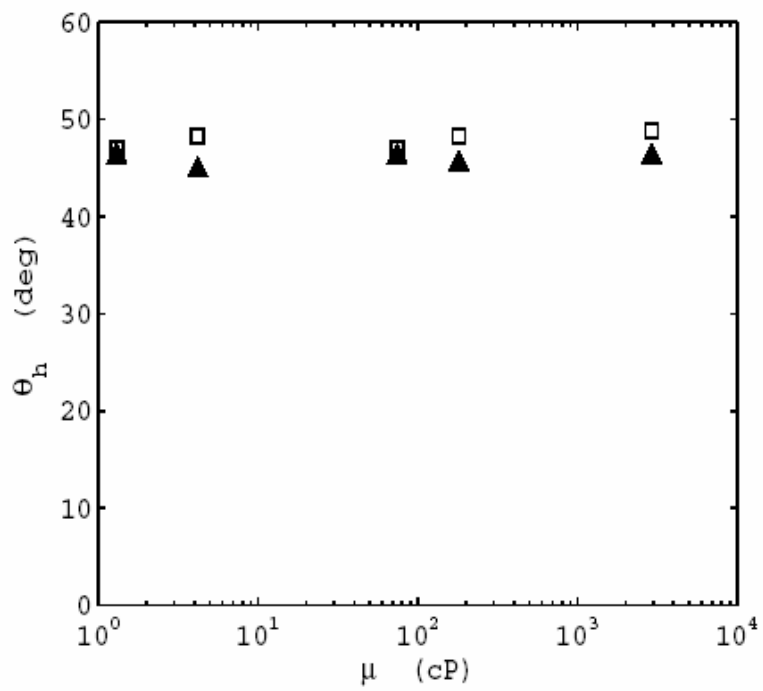


“Local” importance of viscous forces

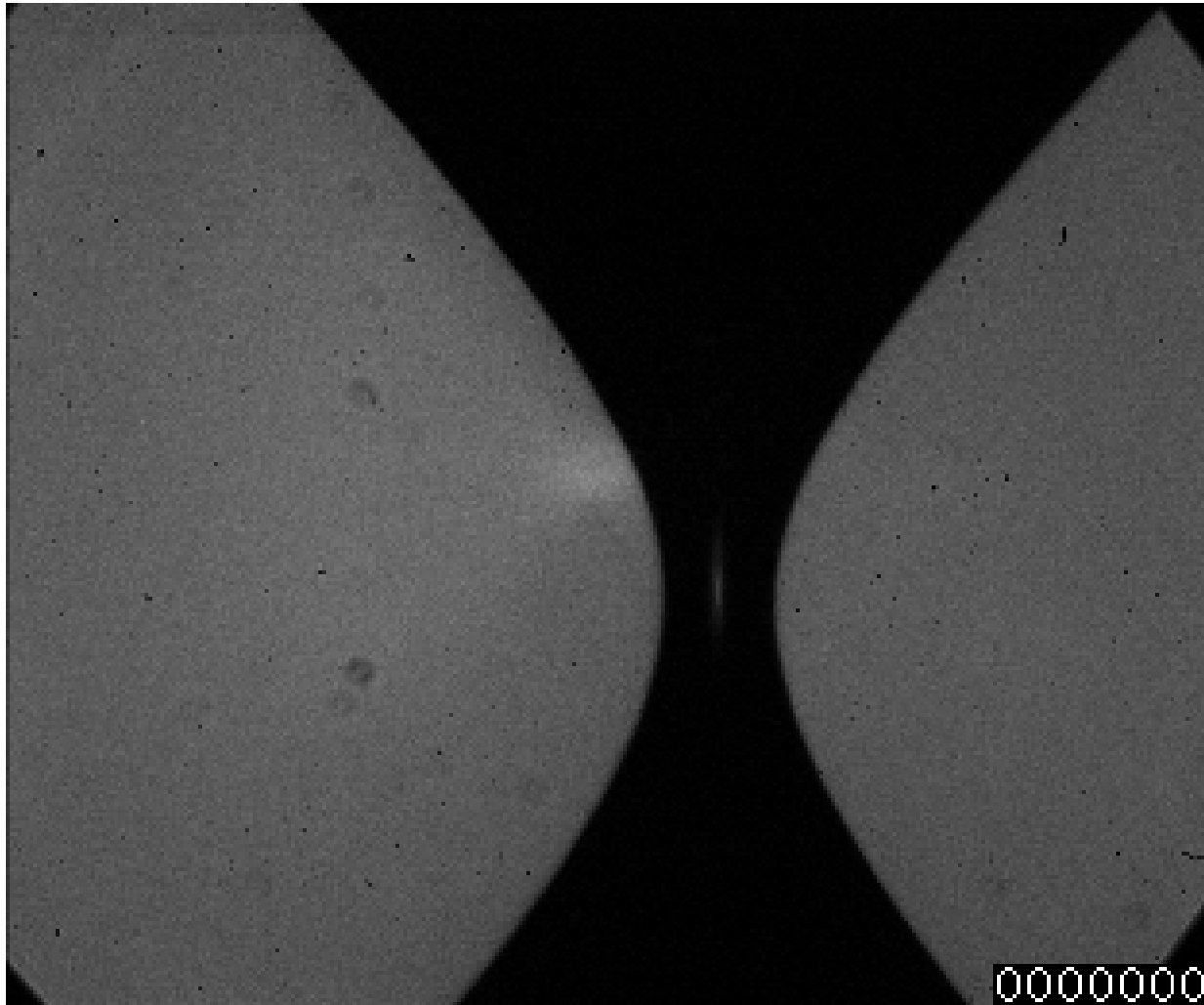




Increasing viscosity



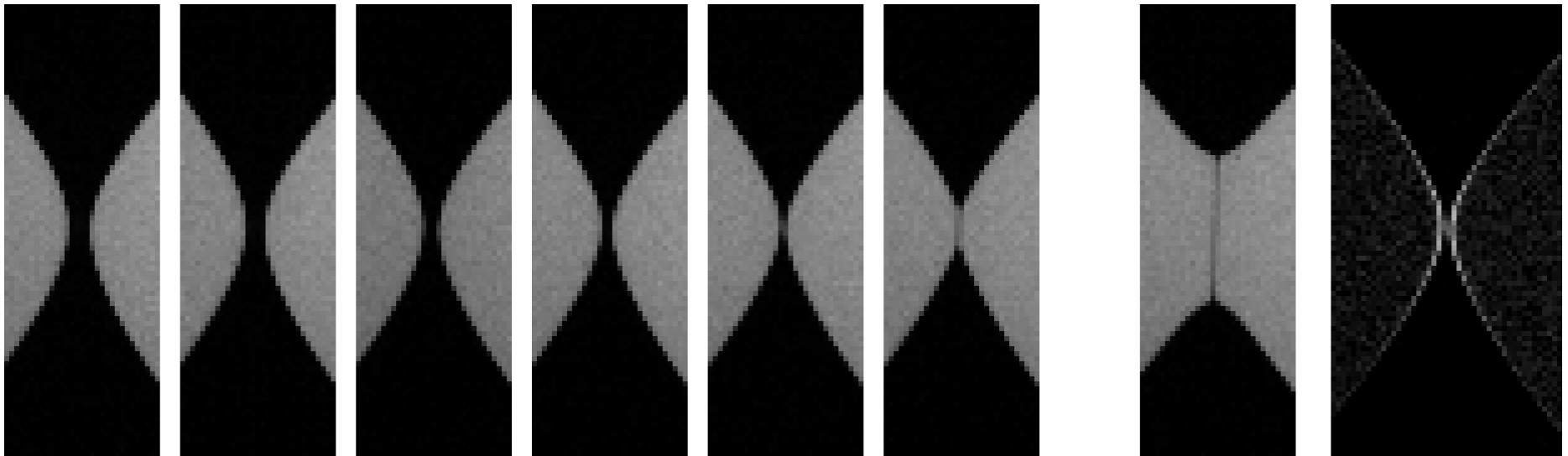
Higher viscosity



100,000 fps

Higher viscosity

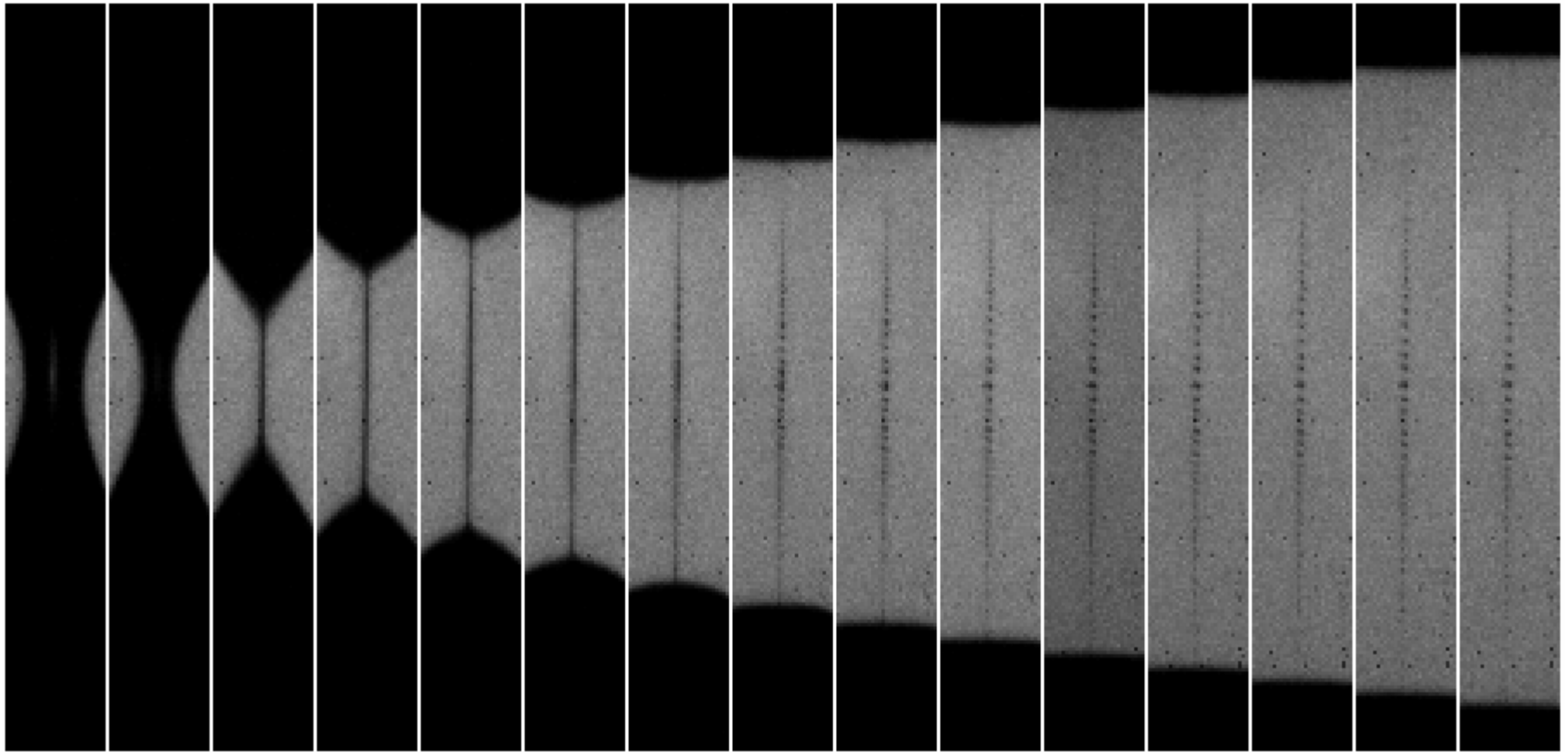
75% glycerin, $\mu = 20 \mu_w$



$dt = 2 \mu s$

$dt = 4 \mu s$

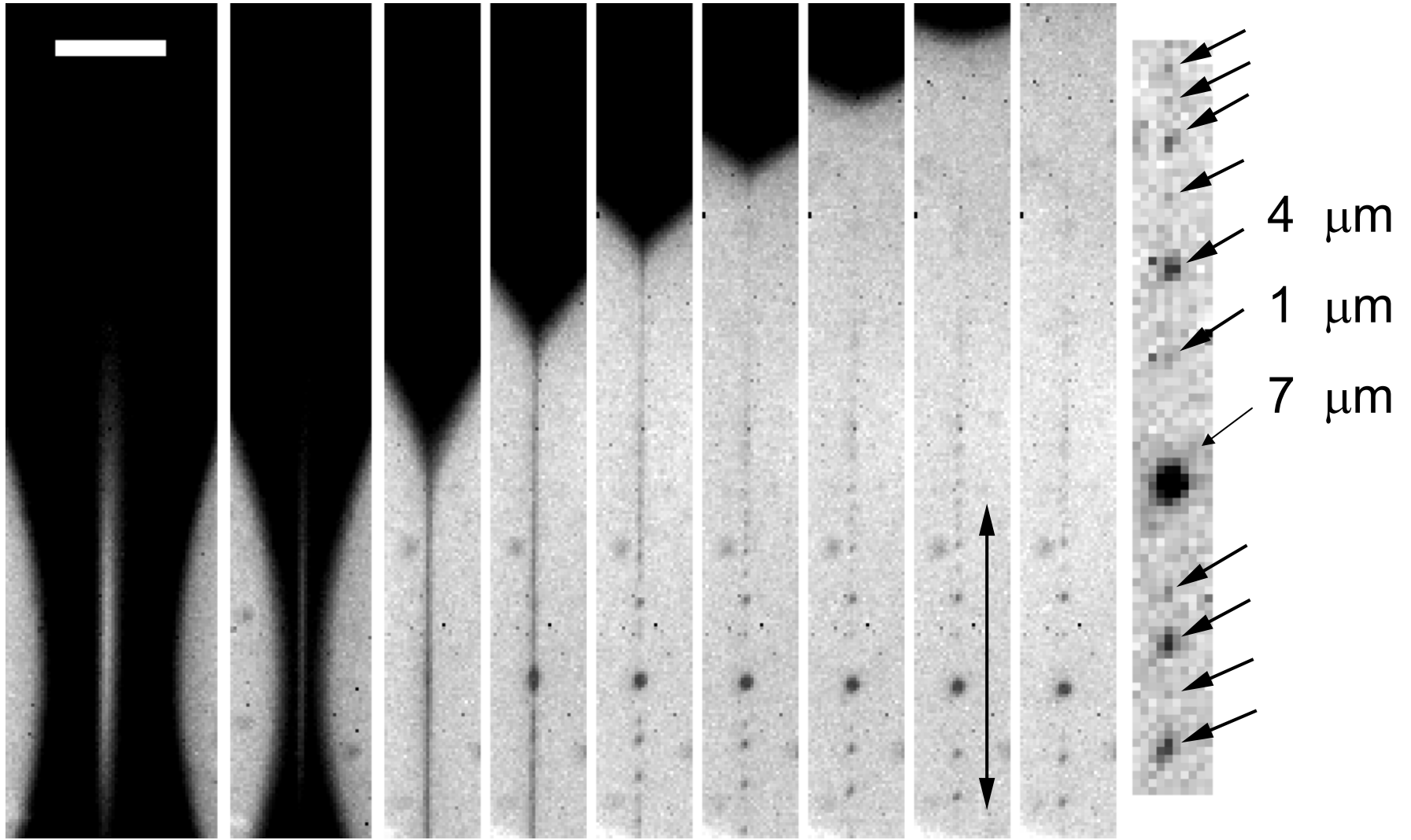
— 50 μm



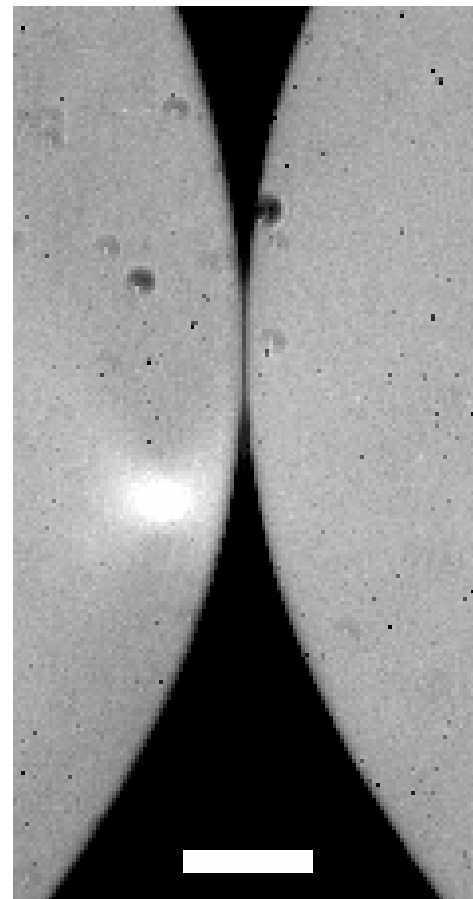
$dt = 10 \mu\text{s}$

50 times viscosity of water

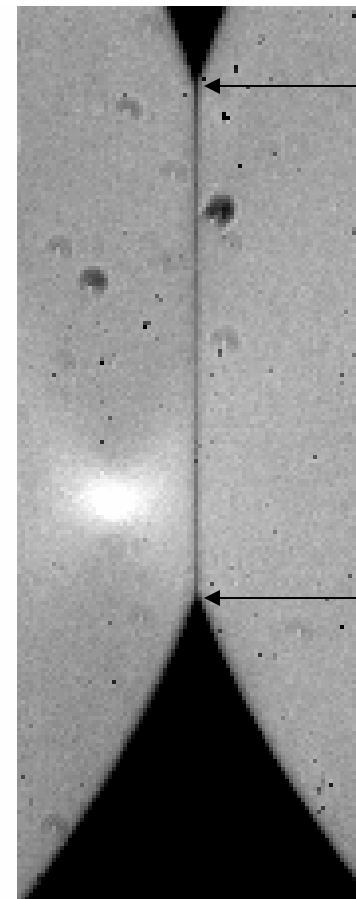
50 μm



99% glycerin
 $\mu = 0.80 \text{ Pa s}$



100 μm



3 μm / px

400 μm

650 μs later

Summary on bubble pinch-off

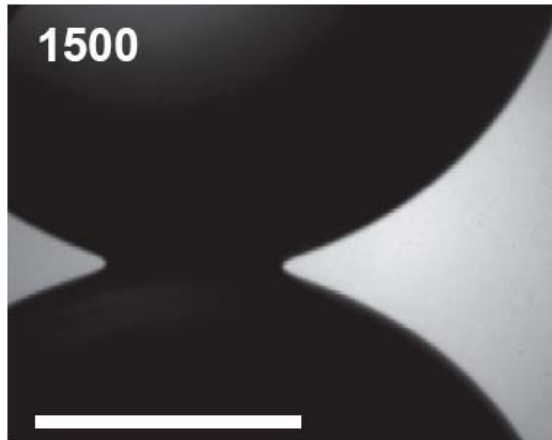
- Most results consistent with *Burton et al. (2005)*
- Exponent $> 1/2$. For water = 0.54 - 0.60
- No 'Rupture' of air tube in water at 50 μm
- For viscous liquid find $\sim 3 \mu\text{m}$ air tube
- No power laws for intermediate viscosities

Outline

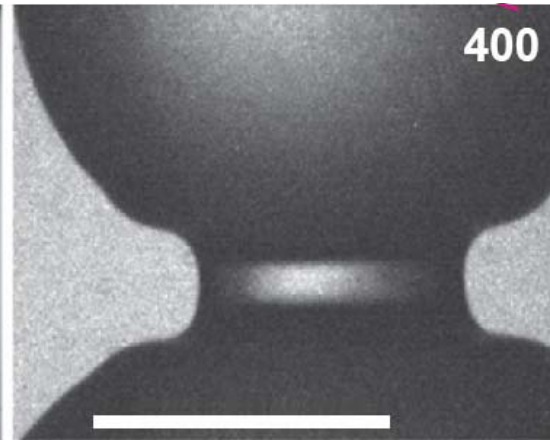
- **1. How fast do free-surface flows move?**
why 1,000,000 fps?
- **2. Imaging and high-speed camera types**
high-speed CCD video cameras
- **3. The pinch-off of a drop or a bubble
from a nozzle**
Different dynamics!
- **4. Coalescence of two drops or bubbles
or Miscible drops**

4. Coalescence of two drops or two bubbles

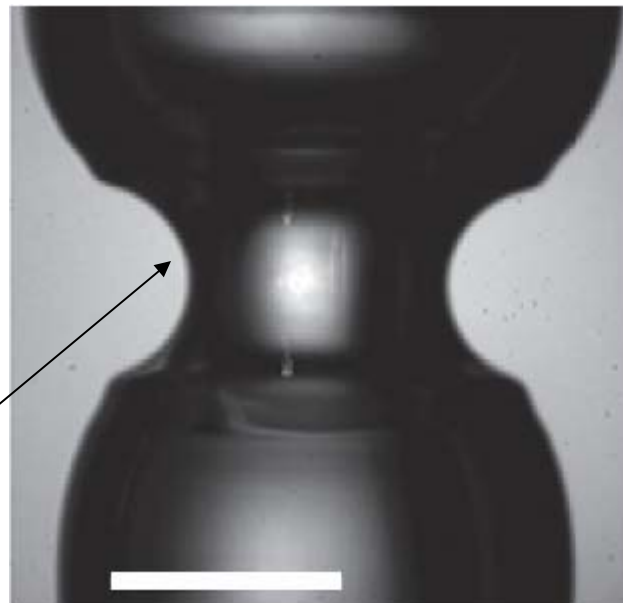
Viscous
Liquids



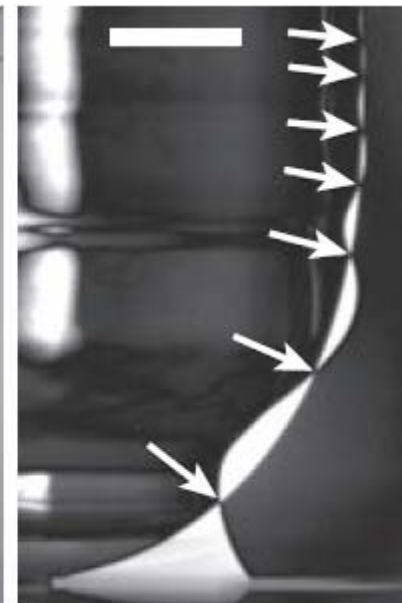
Water
Drops



Bubbles



Marangoni Waves
Water/Etanol



What determines the speed of the coalescence?

Simple Capillary-Inertial Motions Contraction of an air disc

- High surface curvature produces capillary pressure, Young-Laplace
- Inertia or surface tension resists this motion

Surface tension- Capillary pressure

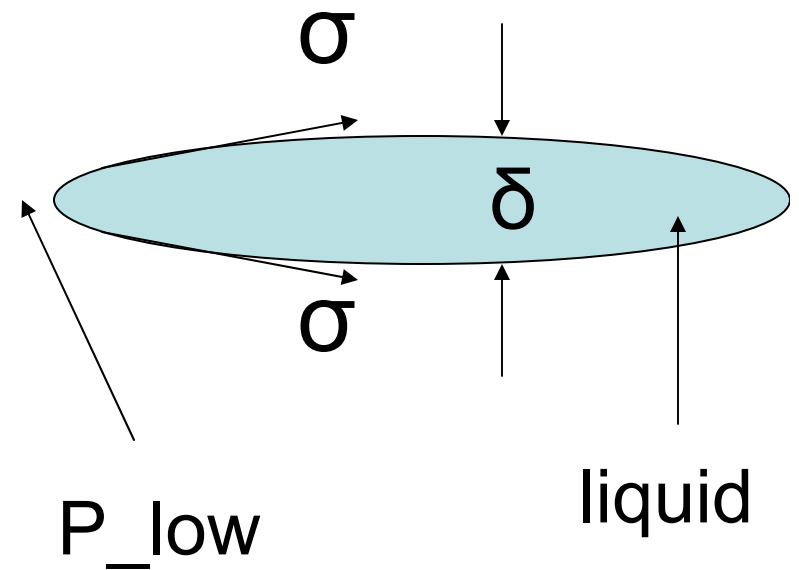
P_{atm}

Young-Laplace law

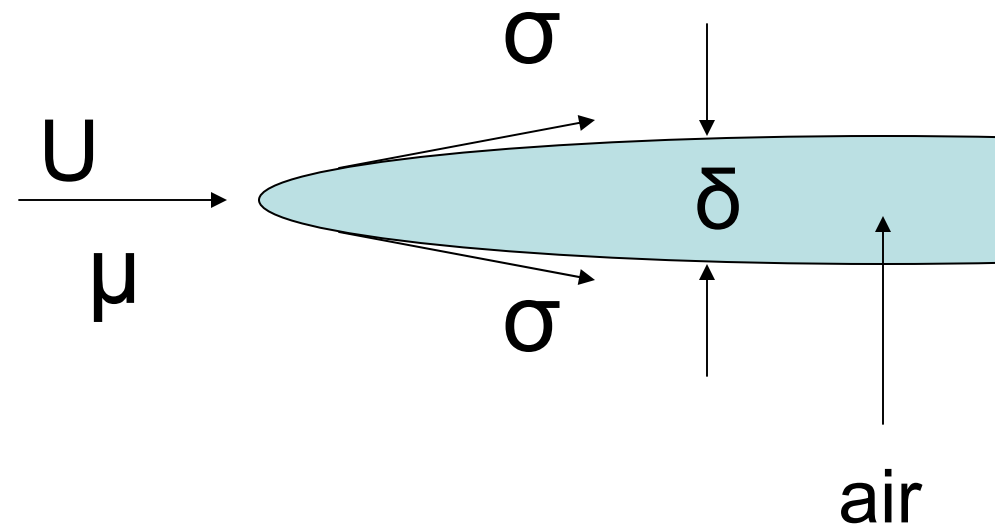
$$\Delta p = \sigma \left(\frac{1}{R_1} + \frac{1}{R_2} \right)$$

1 micron bubble in water

$$\frac{2\sigma}{R} = \frac{2 \times 0.073}{0.5 \times 10^{-6}} = 2.9 \times 10^5 \text{ N/m}^2 \simeq 3 P_{atm}$$



Surface tension-viscous balance

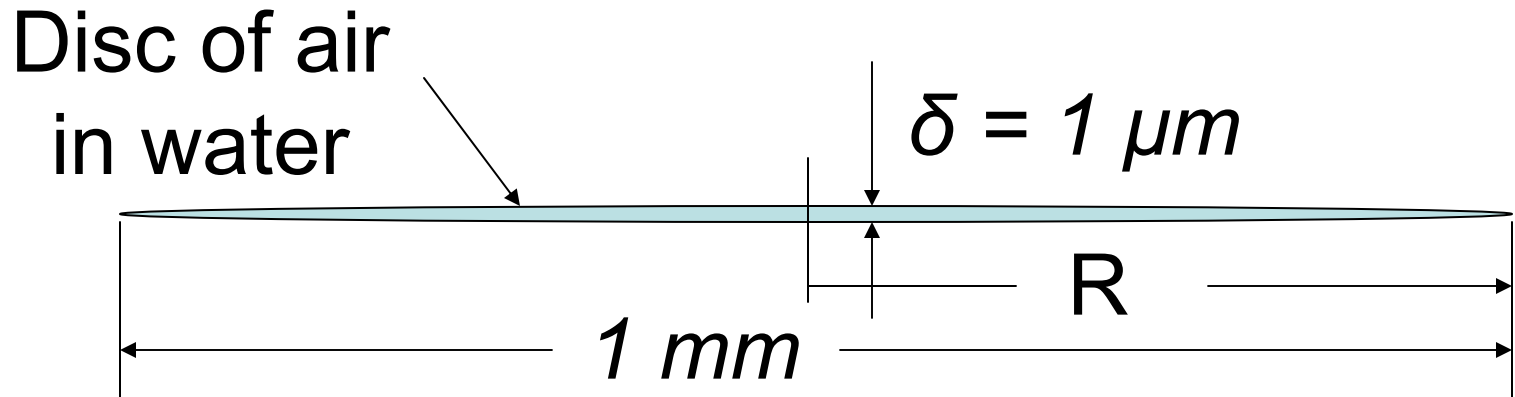


- Viscous force $\sim \mu U / \delta$
- Surface tension $\sim \sigma / \delta$
- Become equal when $\mu U / \delta = \sigma / \delta$

$$U = \sigma / \mu \quad \text{or} \quad Ca = U\mu / \sigma = 1$$

- For water $\sigma = 0.073 \text{ N/m}$
- Dynamic viscosity $\mu = 0.001 \text{ Pa s}$

$$\rightarrow \underline{U = 73 \text{ m/s!}}$$



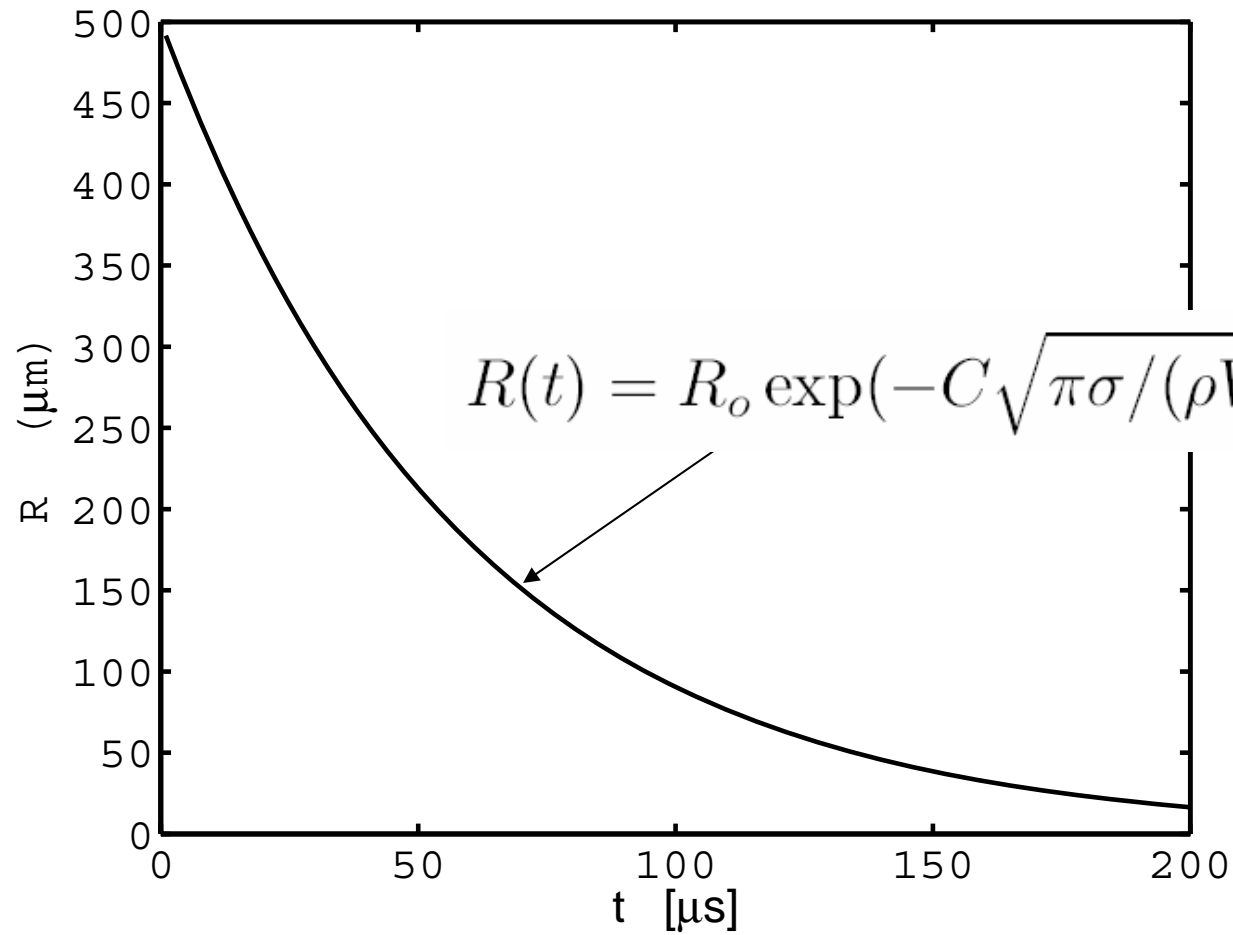
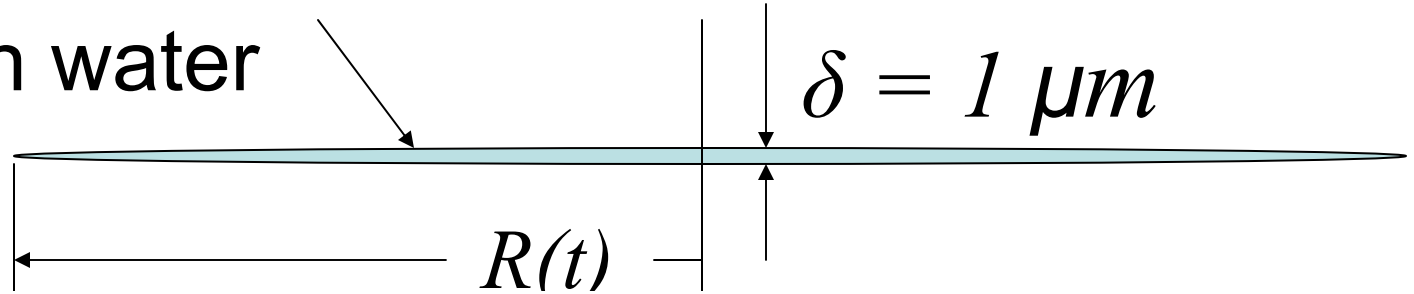
- Including inertia, dynamic pressure $\sim \rho U^2$
- Capillary pressure $\sim \sigma / \delta$
- In balance $U \sim C (\sigma / \rho \delta)^{1/2}$
- Disc thickness: $\delta \pi R^2 = Vol. \Rightarrow \delta = Vol. / \pi R^2$

$$U = -dR/dt = (\sigma / \rho \delta)^{1/2} = (\sigma \pi R^2 / \rho Vol.)^{1/2}$$

$$-dR/dt = (\sigma \pi / \rho Vol.)^{1/2} R$$

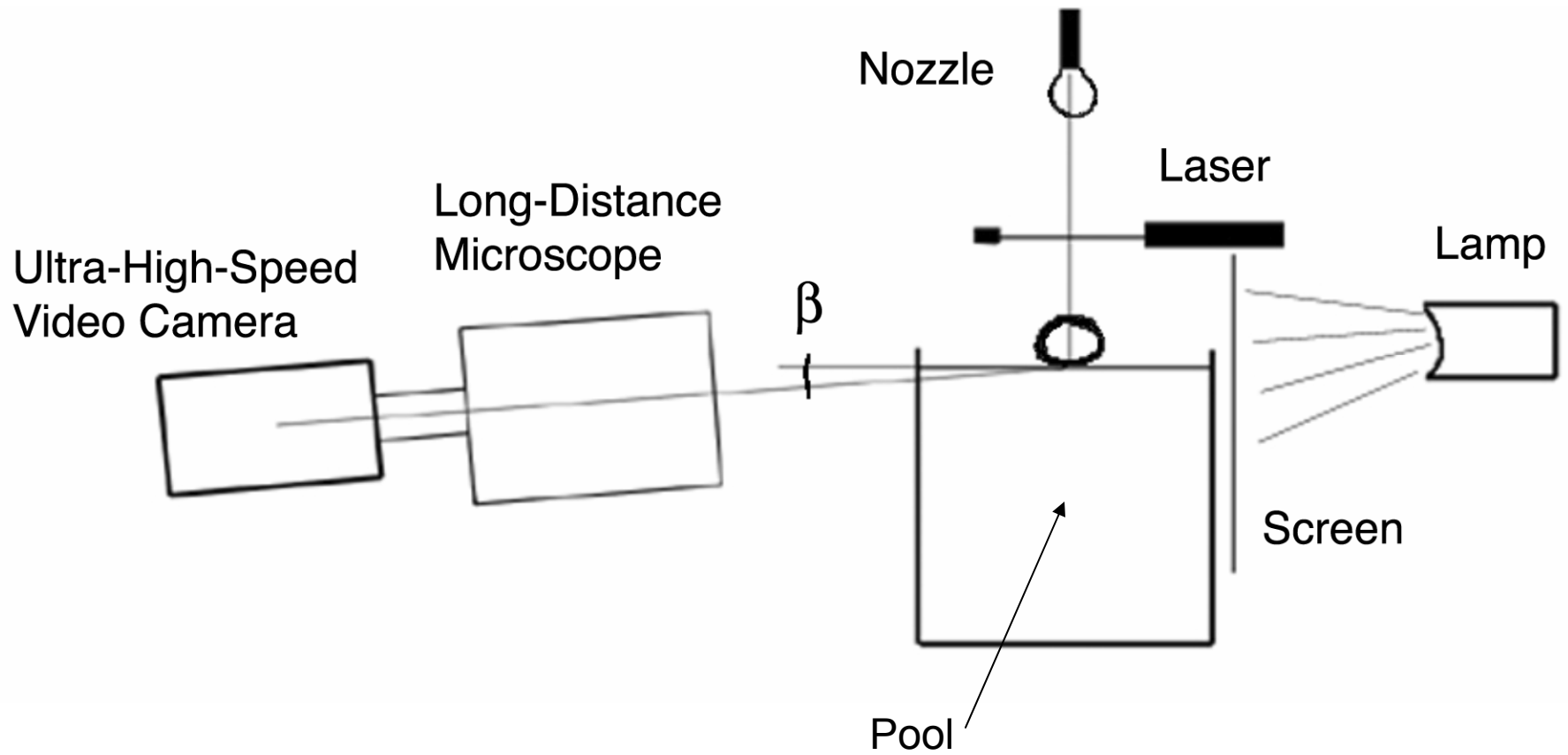
$$\underline{R = R_0 \exp[-(\sigma \pi / \rho Vol.)^{1/2}] t}$$

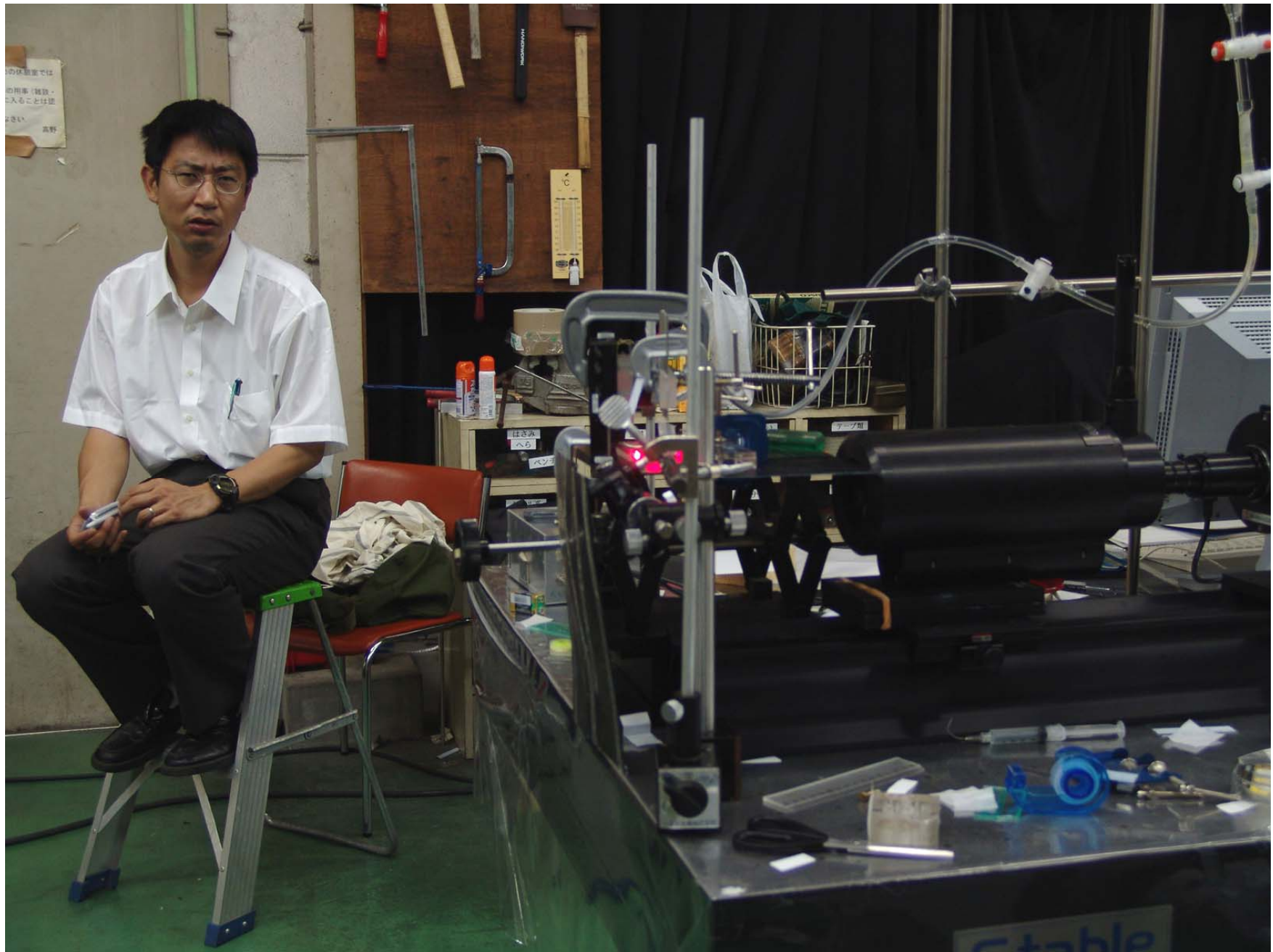
Disc of air
in water

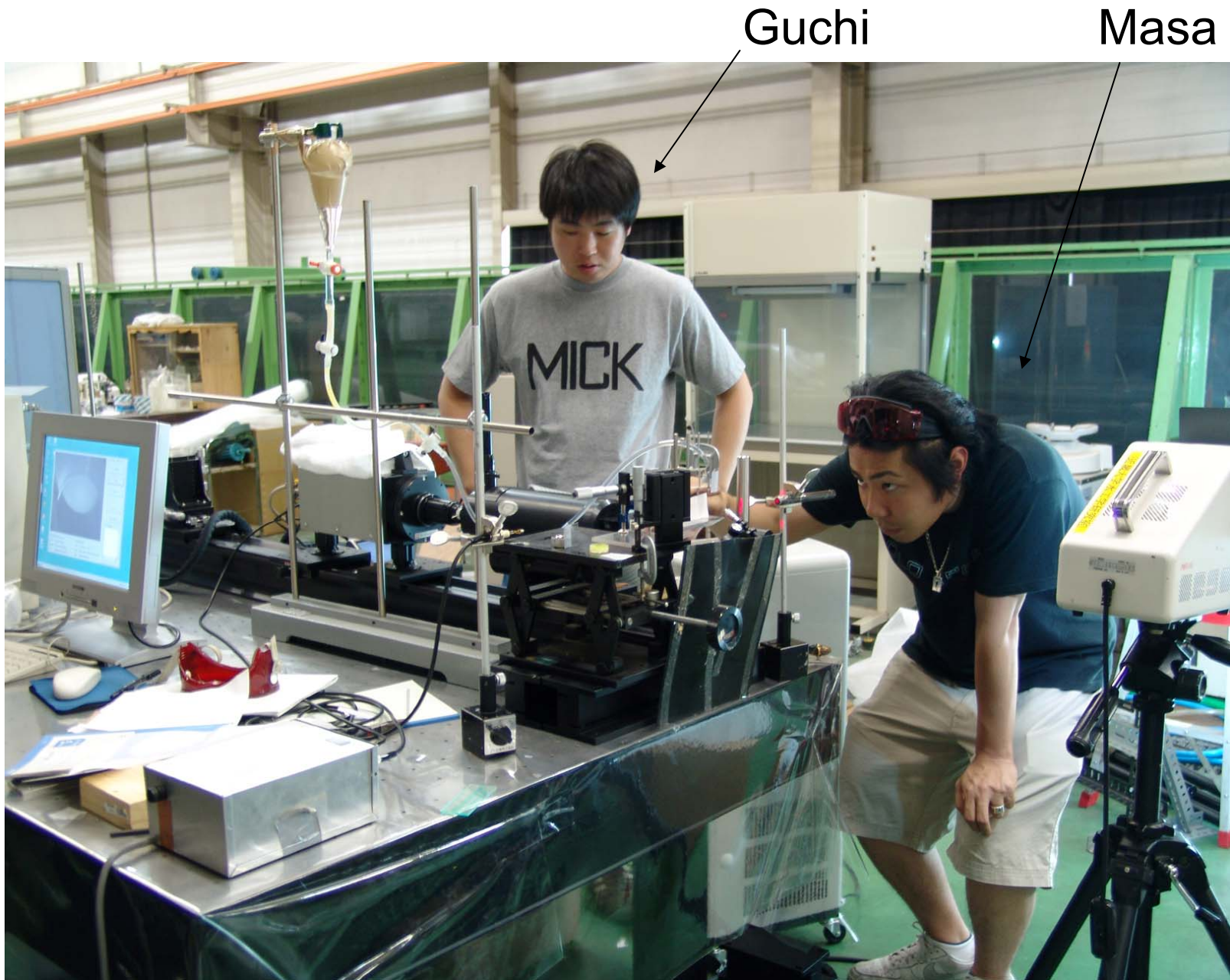


$\Rightarrow 50\,000$ fps

Experimental Setup



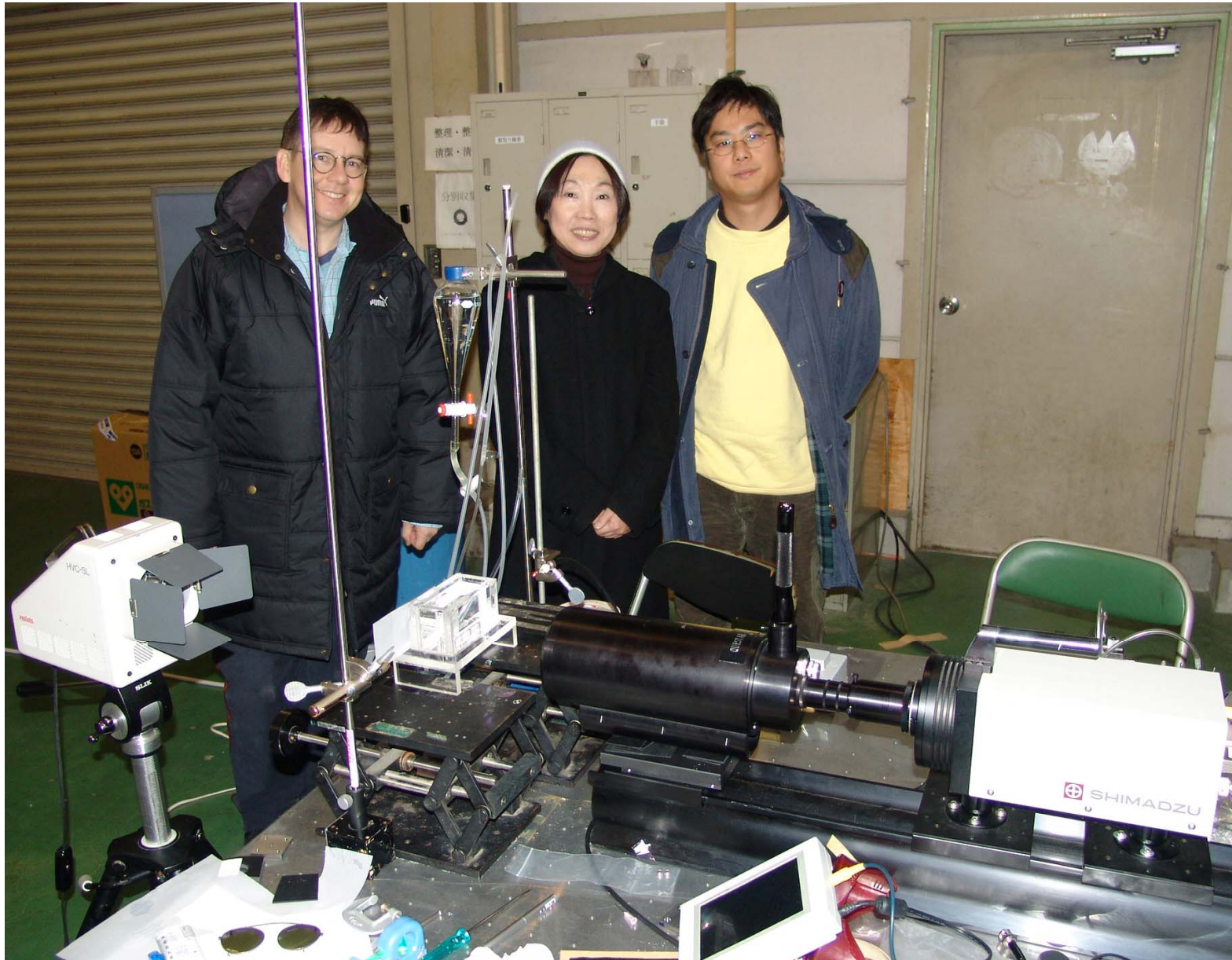




Guchi

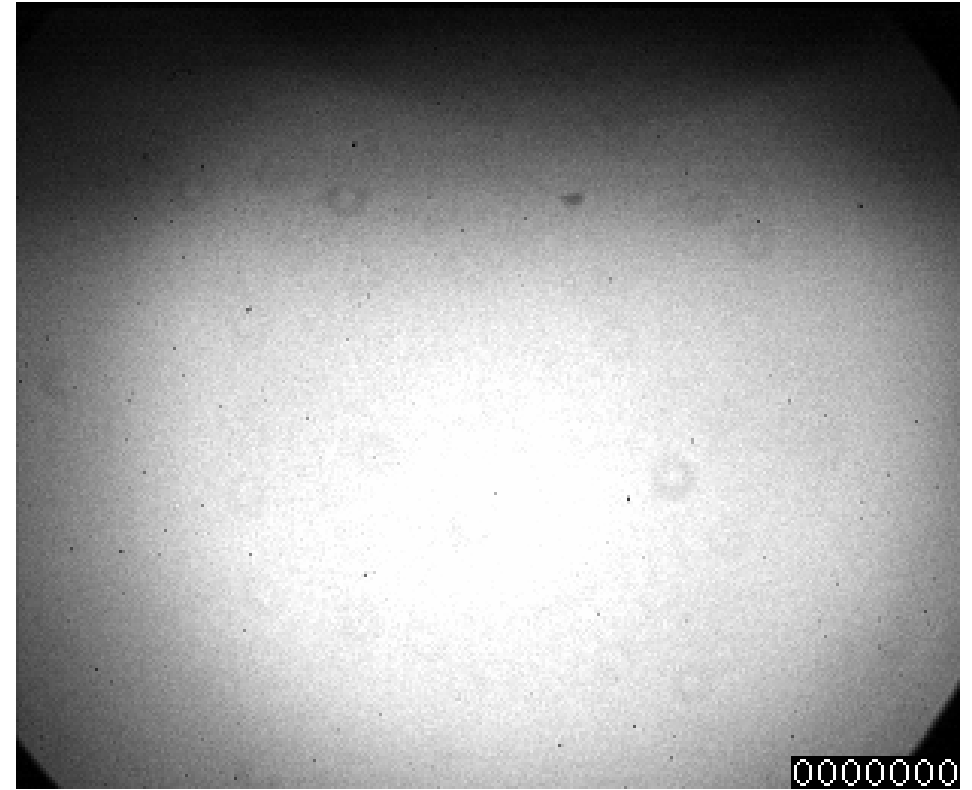
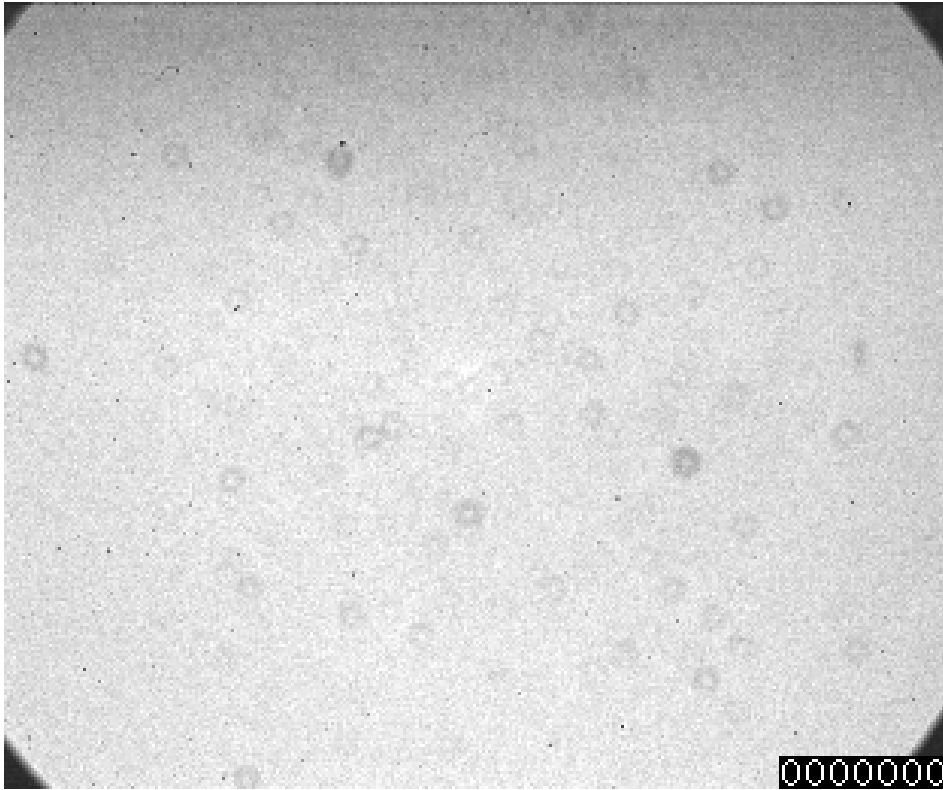
Masa

Osaka, July, 35°C

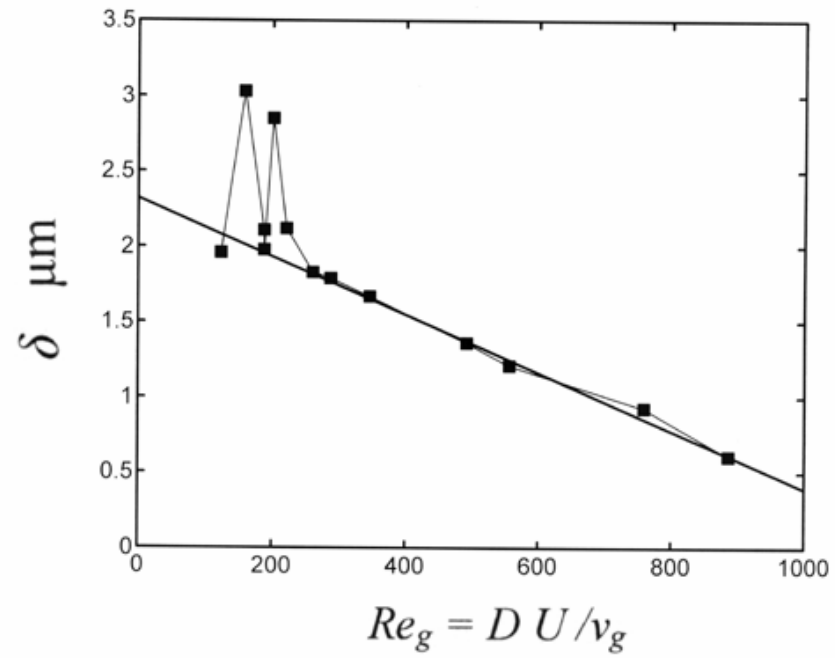
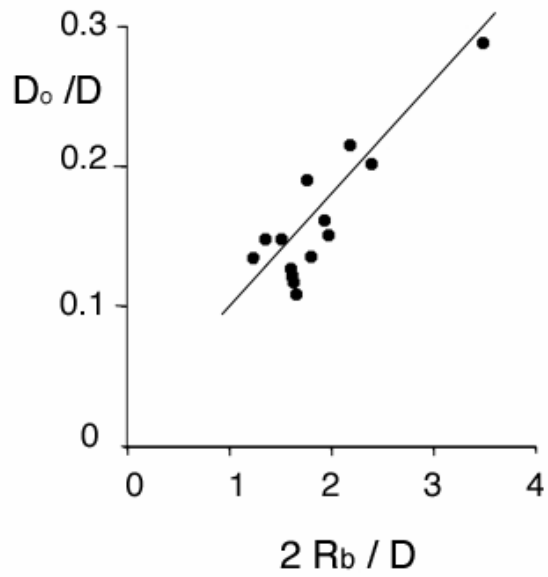
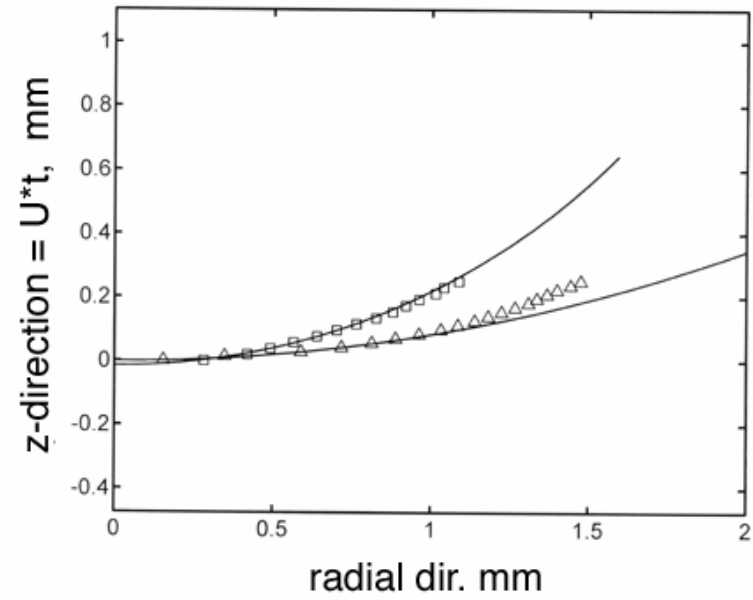
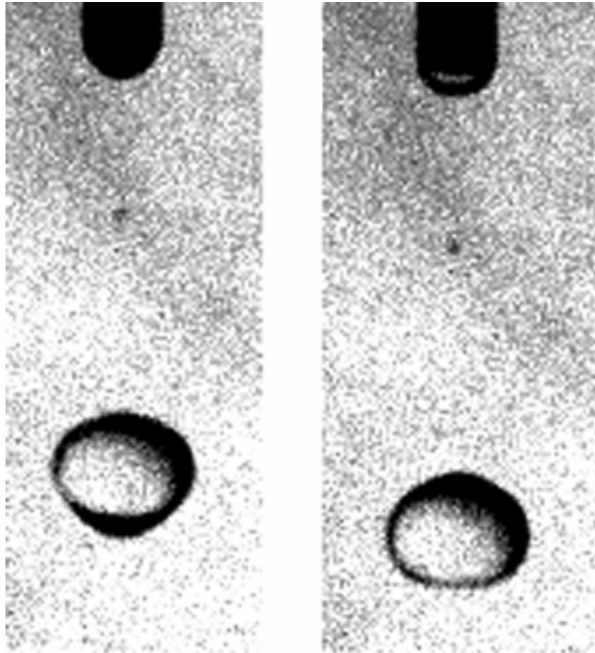


Civil Engineering, Kinki University, Osaka, Japan, 5° C

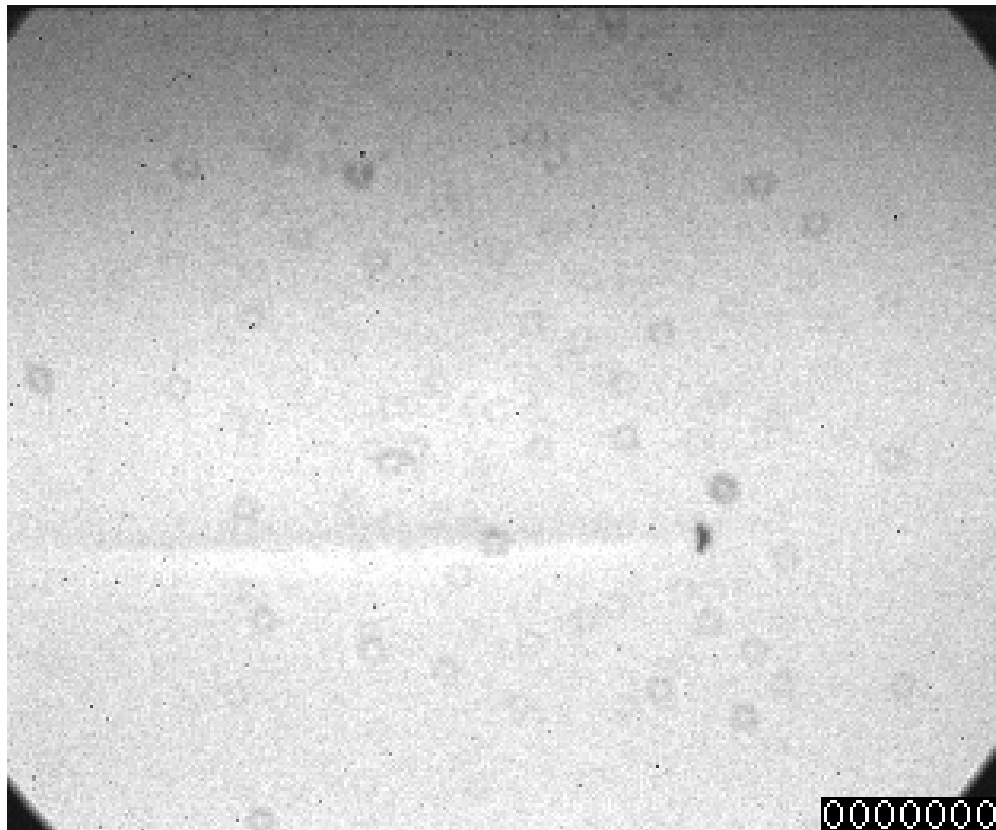
Air entrapment



Bubbles 30 – 50 μm

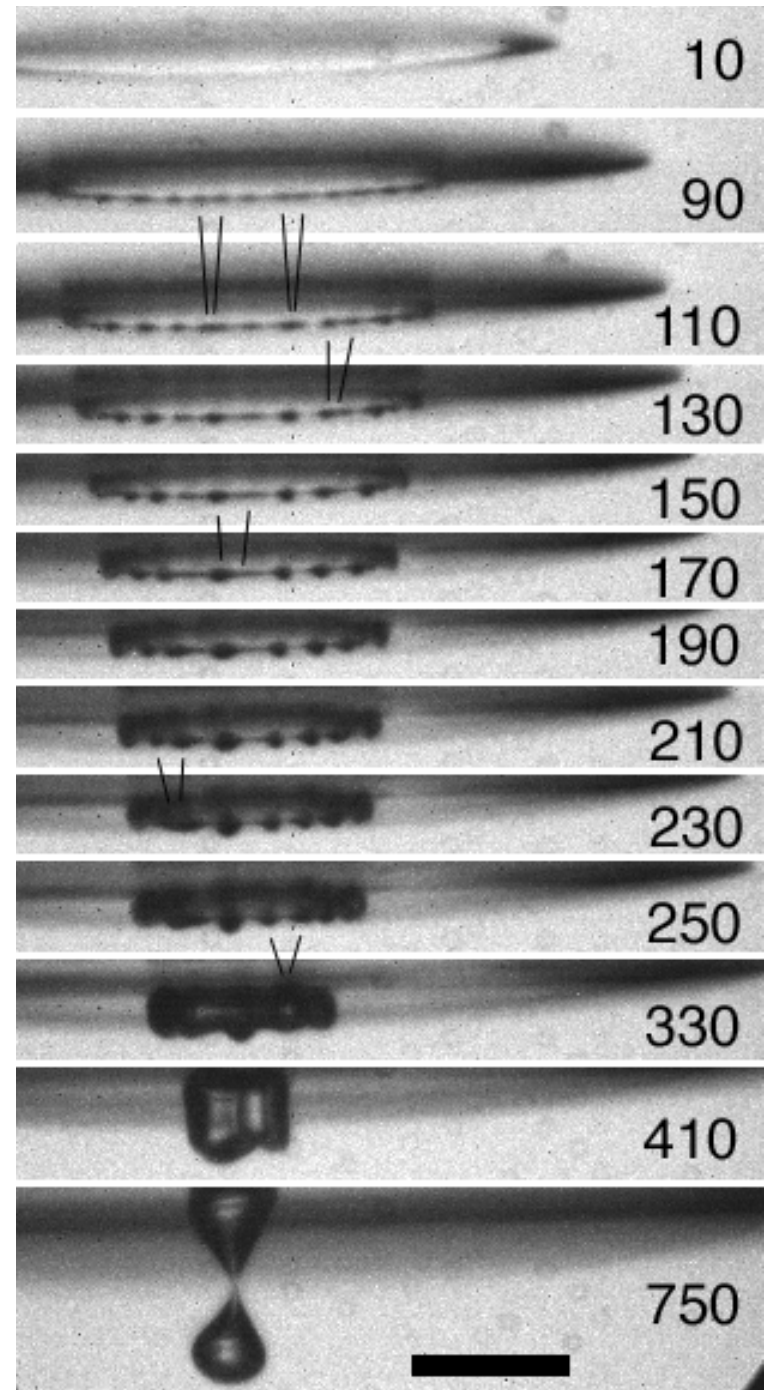


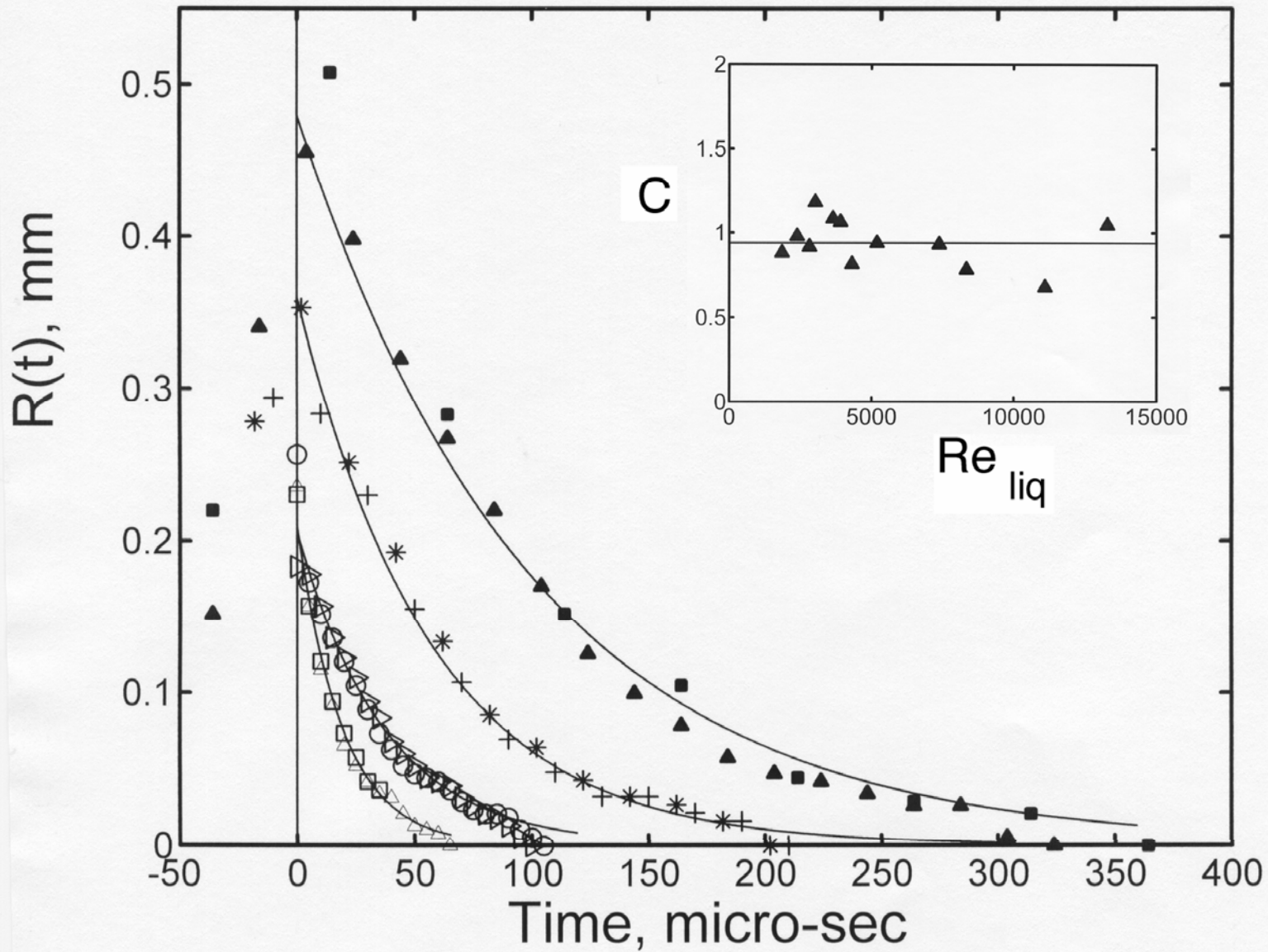
Thoroddsen, Etoh & Takehara,
JFM, 478 (2003)



$\mu\text{s} \rightarrow$

$500 \mu\text{m} \rightarrow$

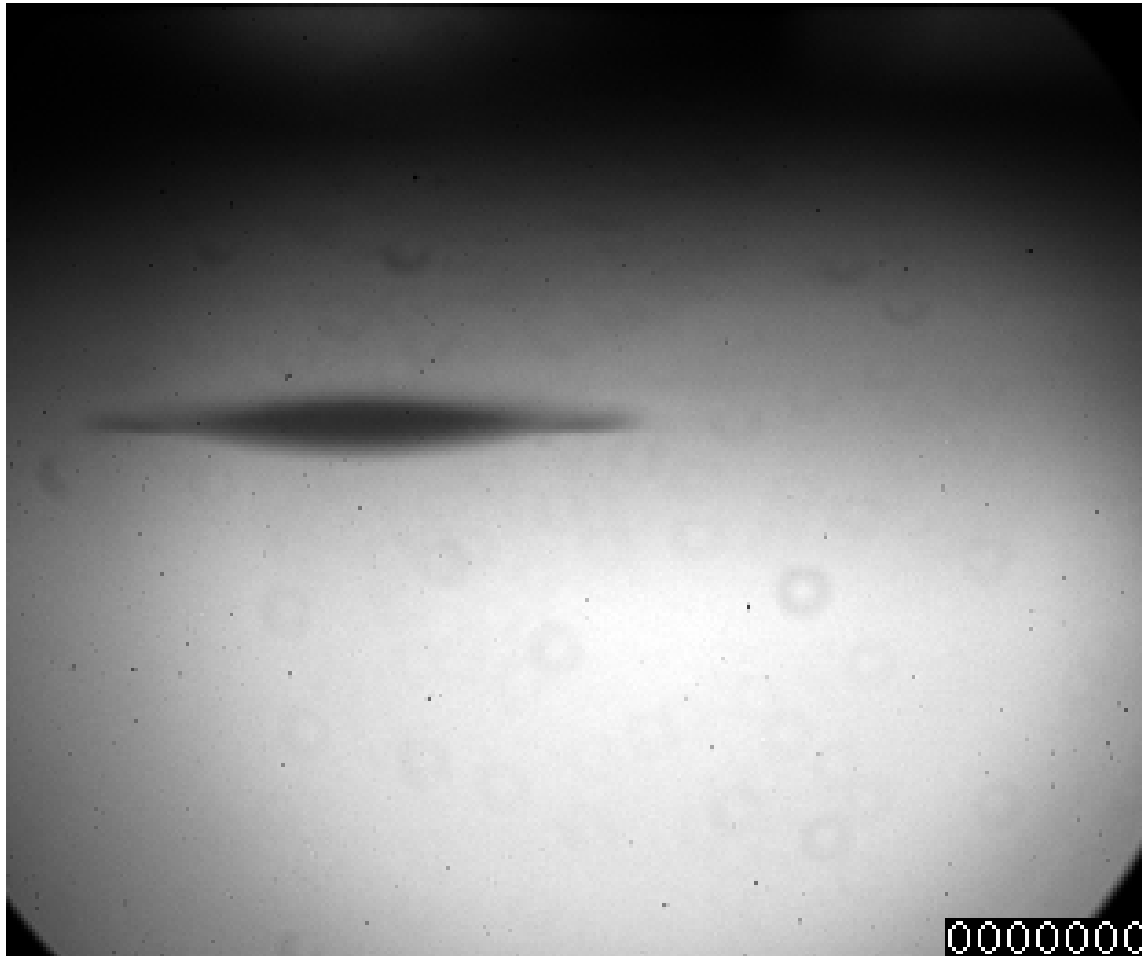




$C \approx 0.94$

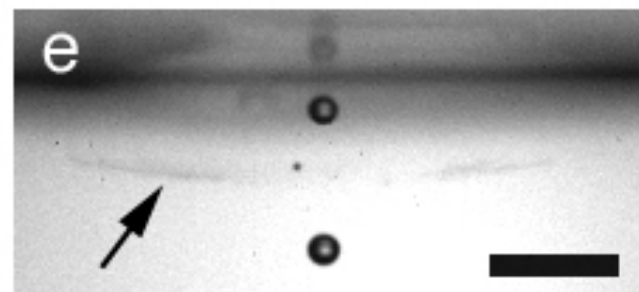
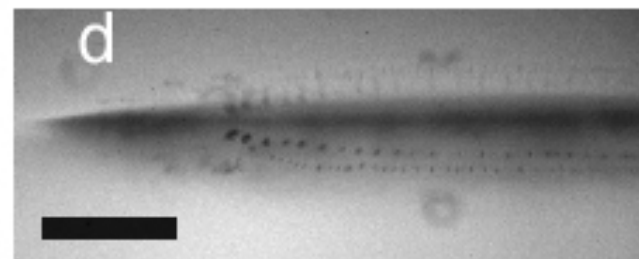
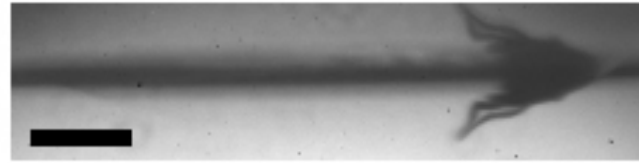
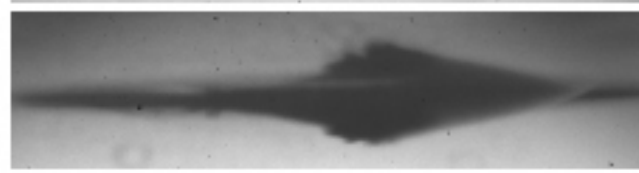
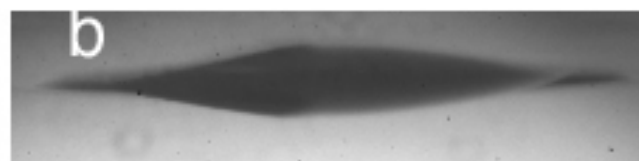
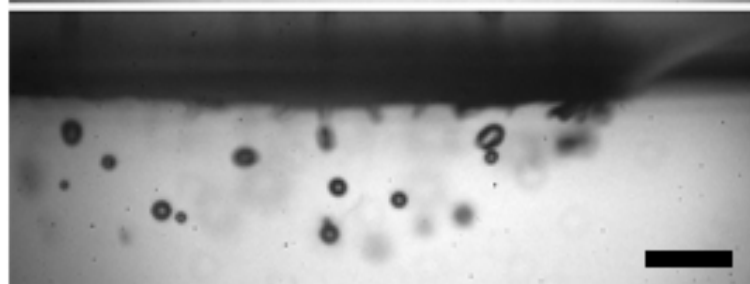
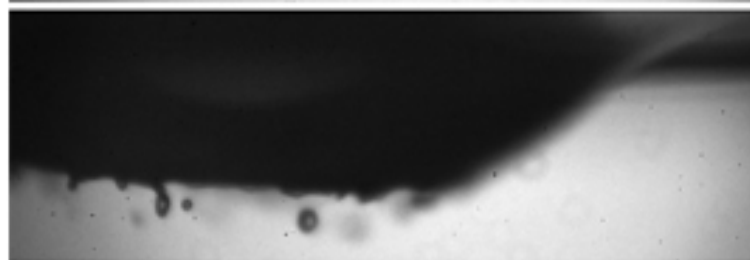
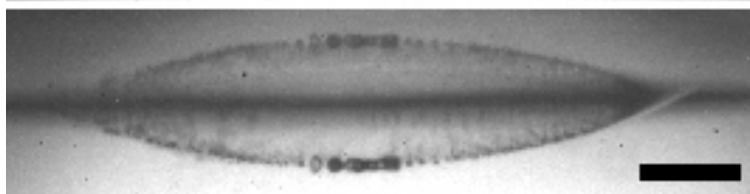
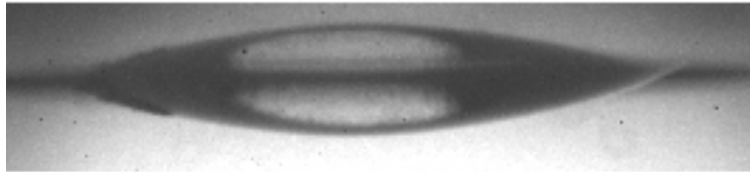
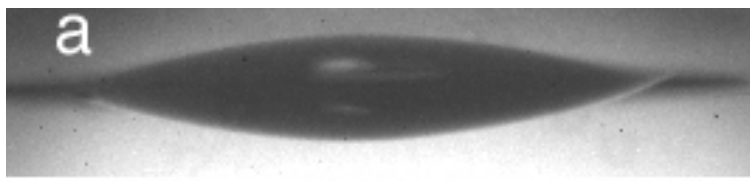
$$R(t) = R_o \exp(-C \sqrt{\pi \sigma / (\rho V ol.)} t)$$

Thin film under drop



Low Re / We

Bars
0.5 mm

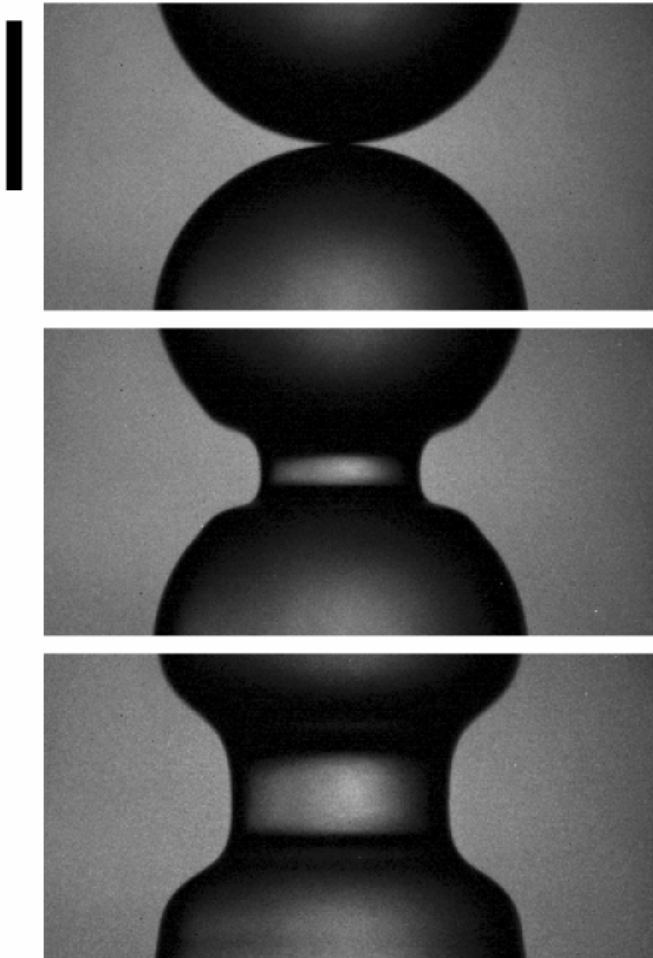


Coalescence of Miscible Drops

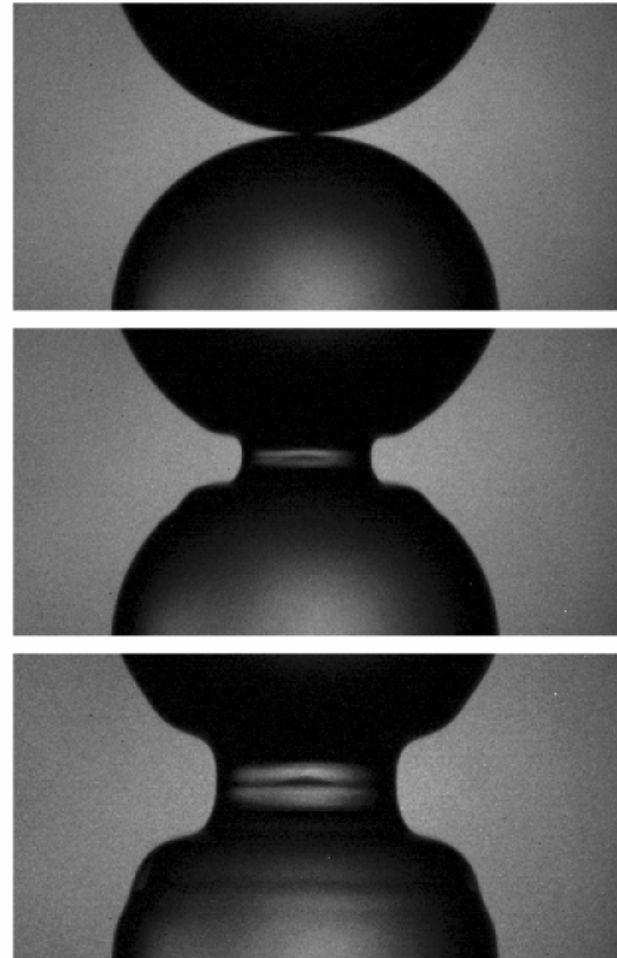
- Water drop coalescing with an ethanol drop (lower σ)

What controls the speed of coalescence?

Thoroddsen, Qian, Etoh & Takehara, (2007)
The initial coalescence of miscible drops,
Phys. Fluids, **19**, 072110



Water on water



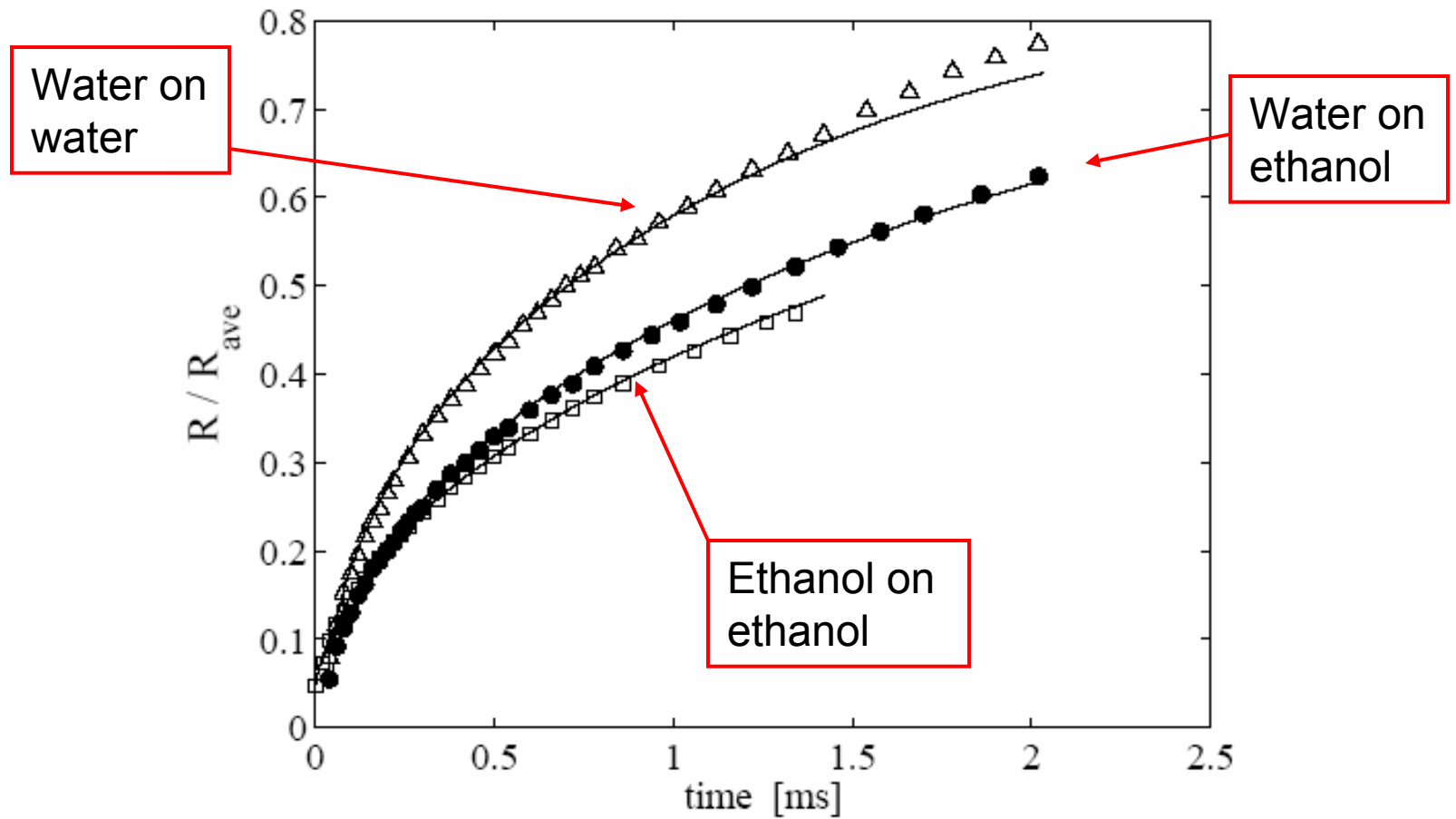
Ethanol on top

Water on bottom

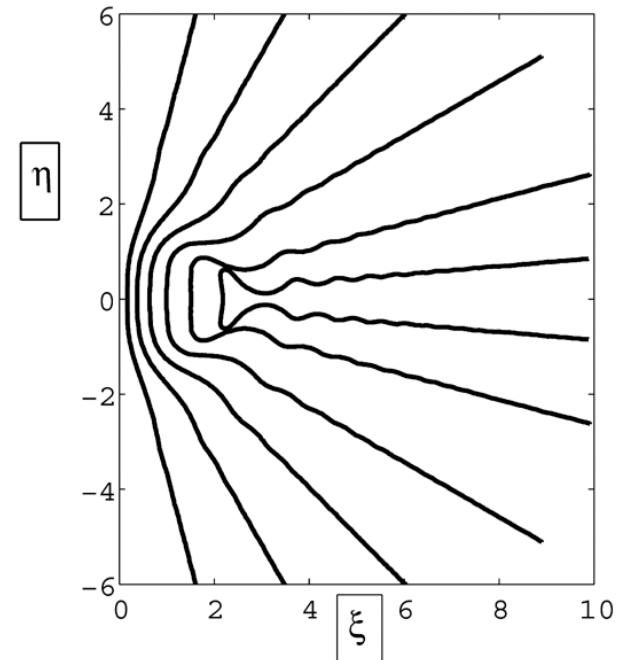
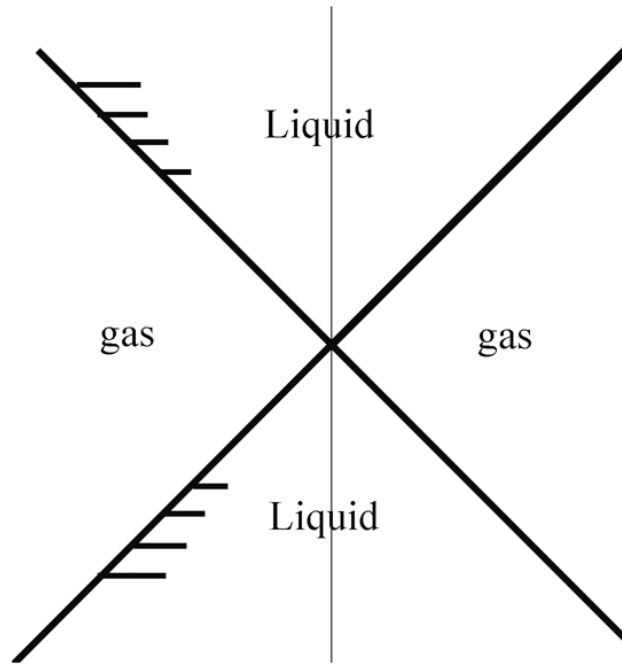
1 ms

Speed of coalescence

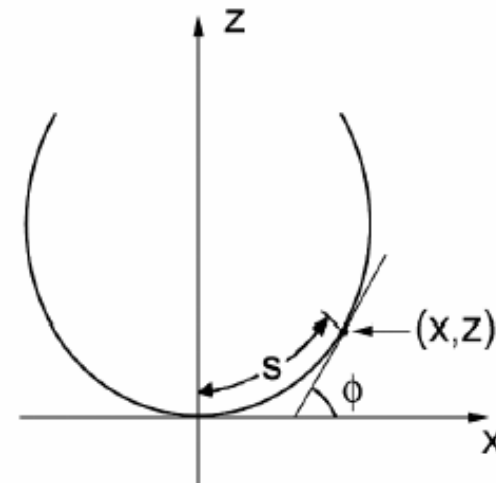
Is determined by the lower surface tension



Selfsimilar Marangoni waves



Wedges or Cones
Self-similar geometry



Water drop coalescing with an ethanol pool

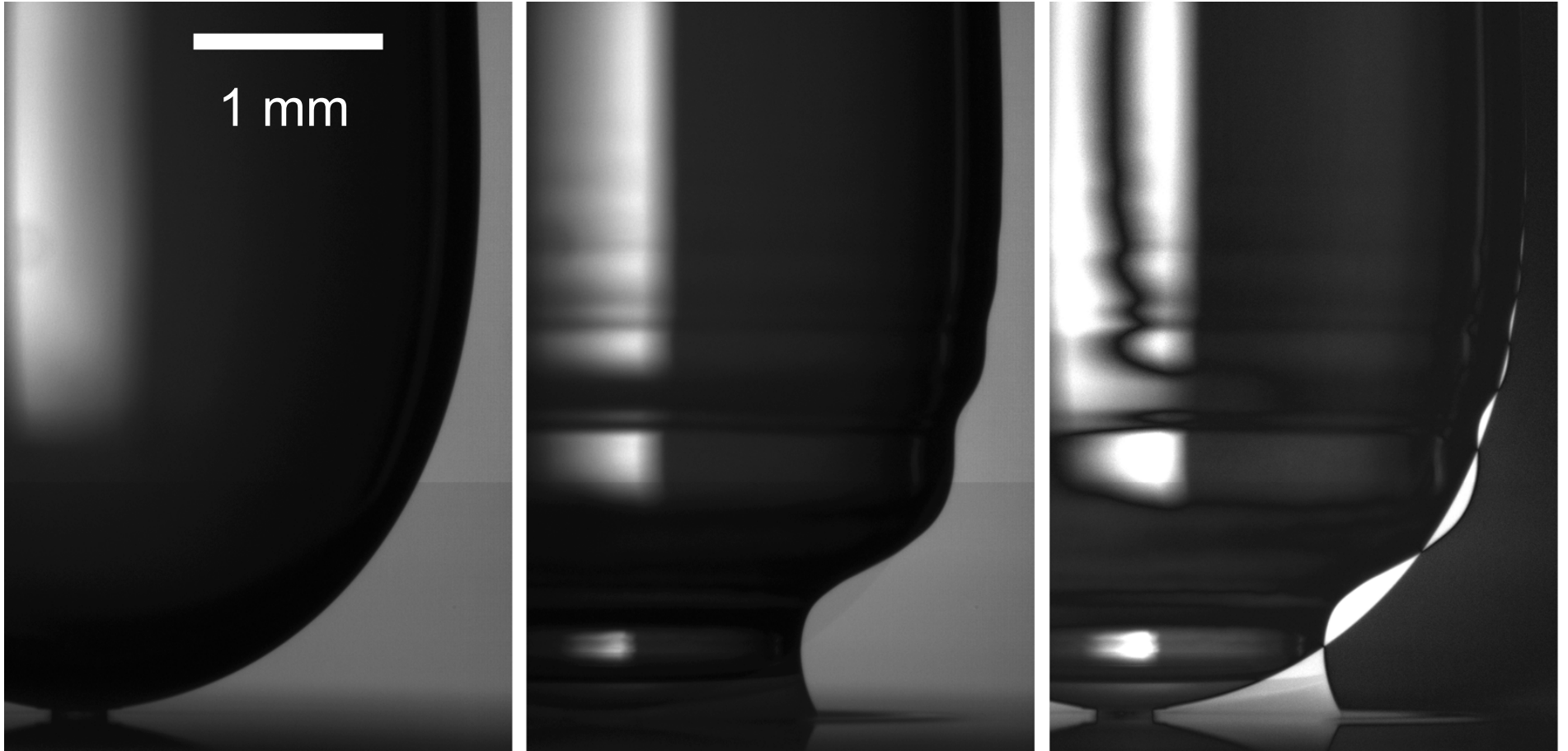
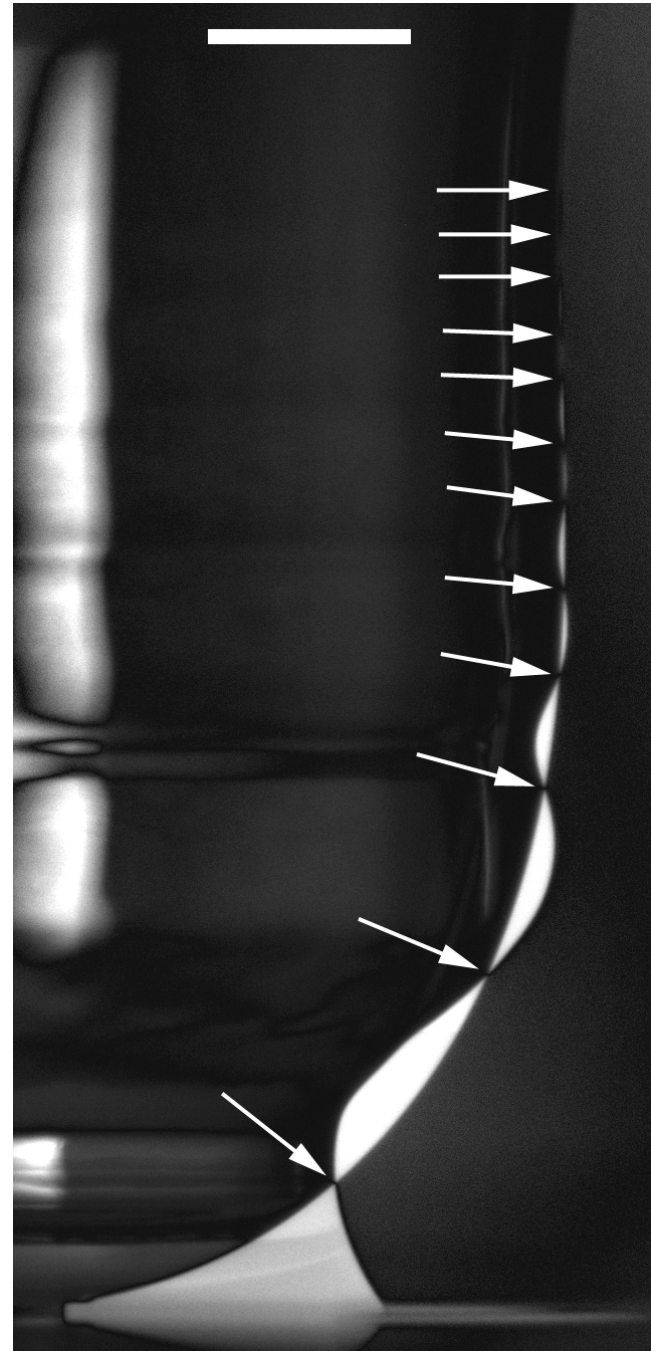
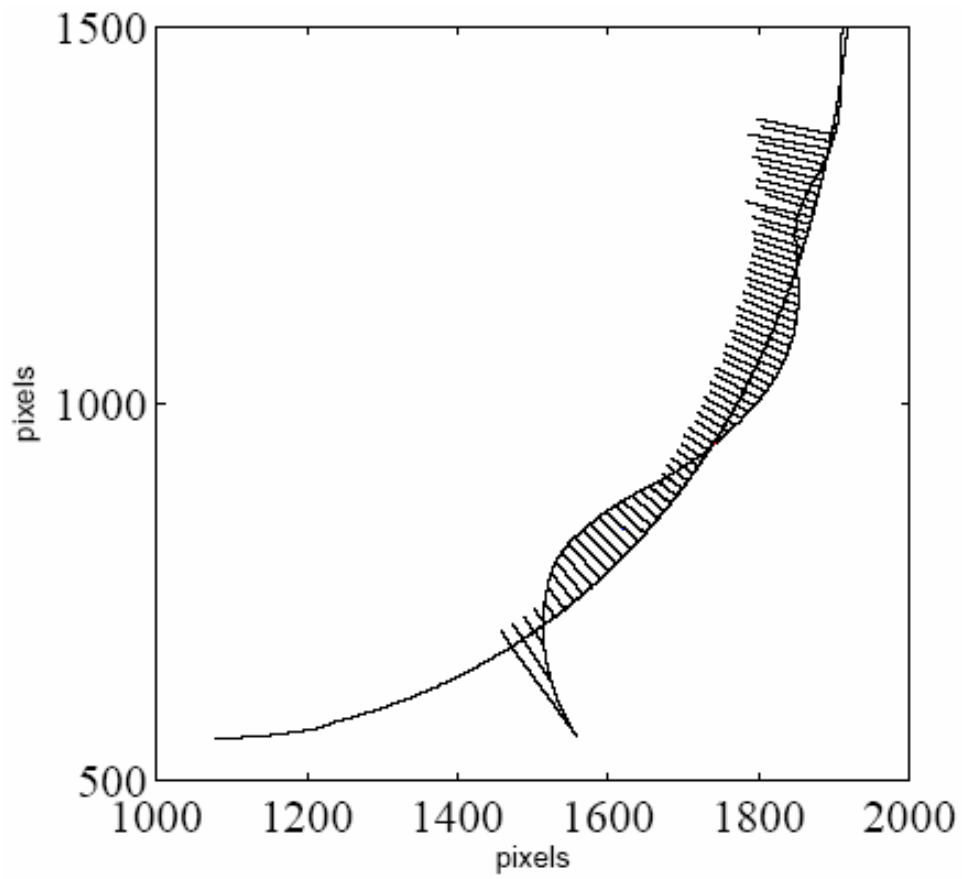
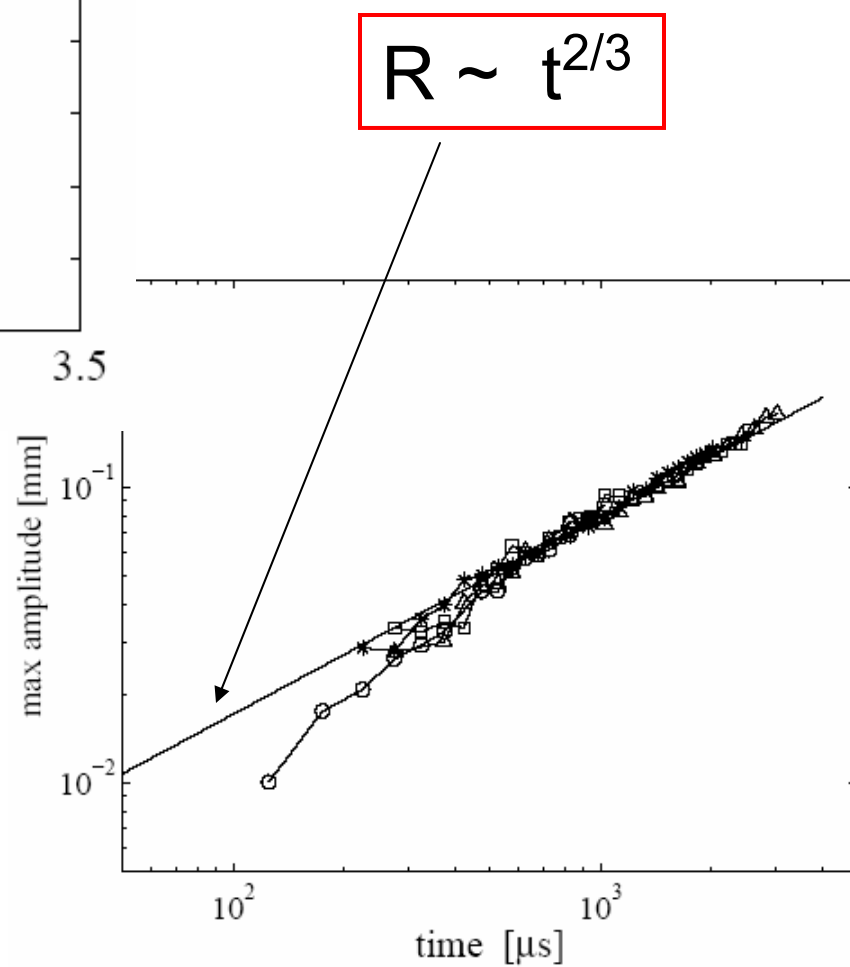
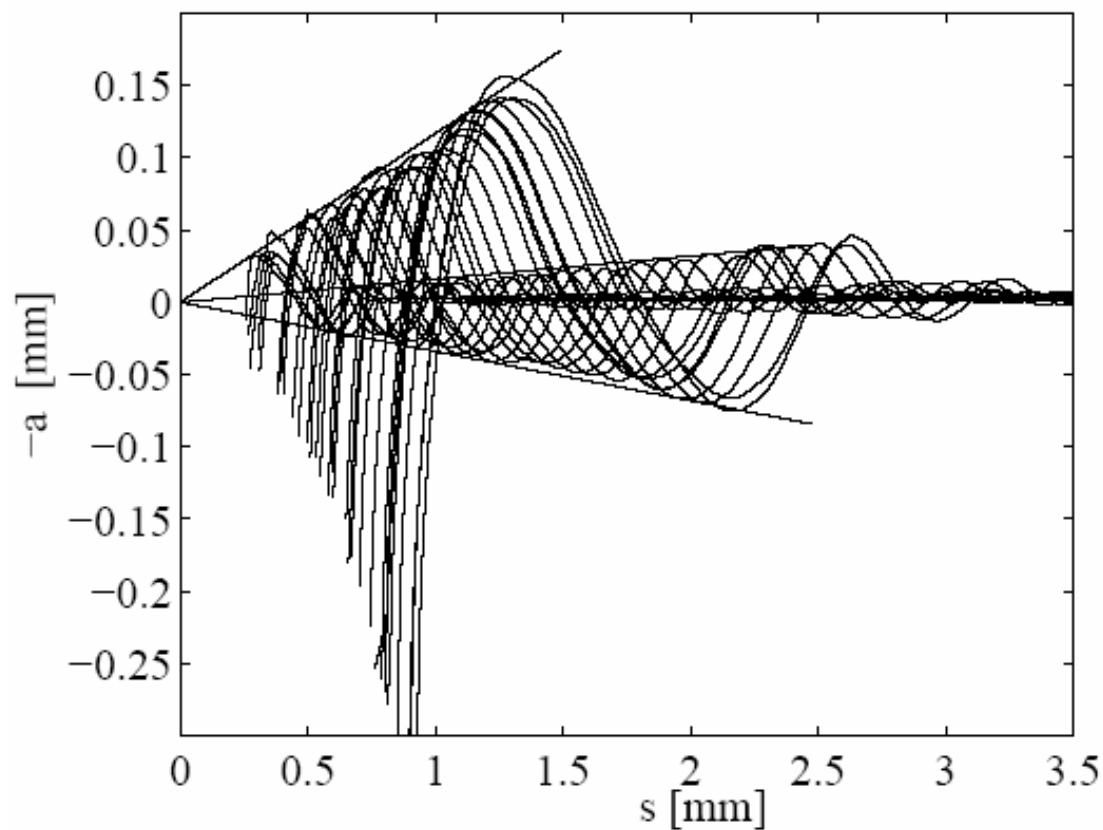
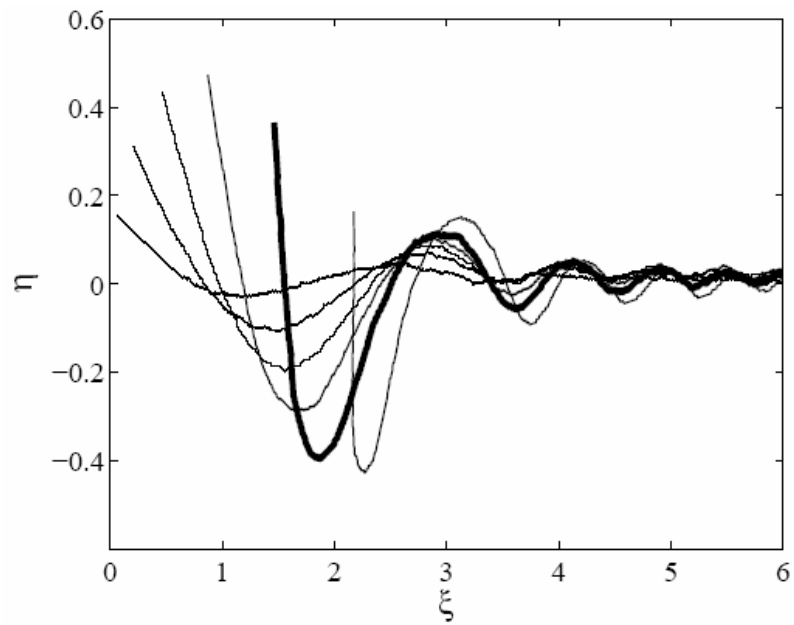
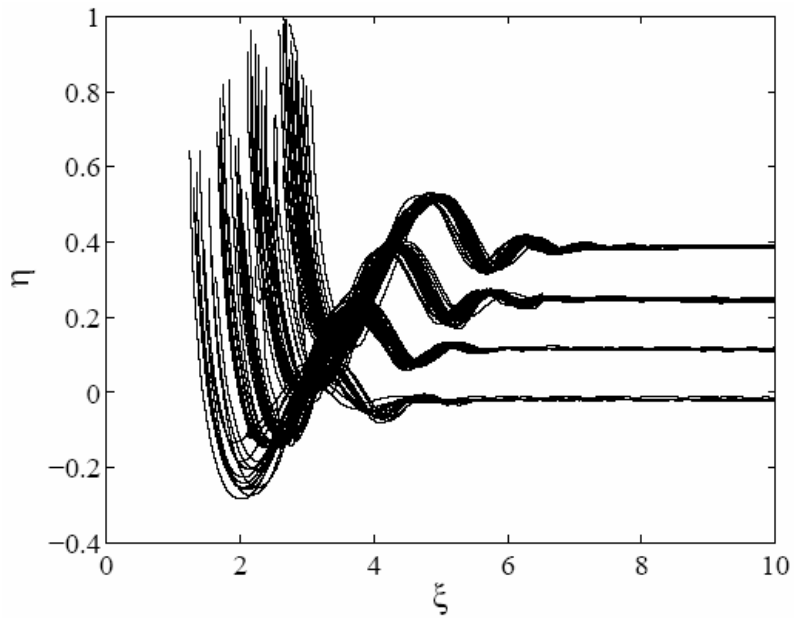


Image difference

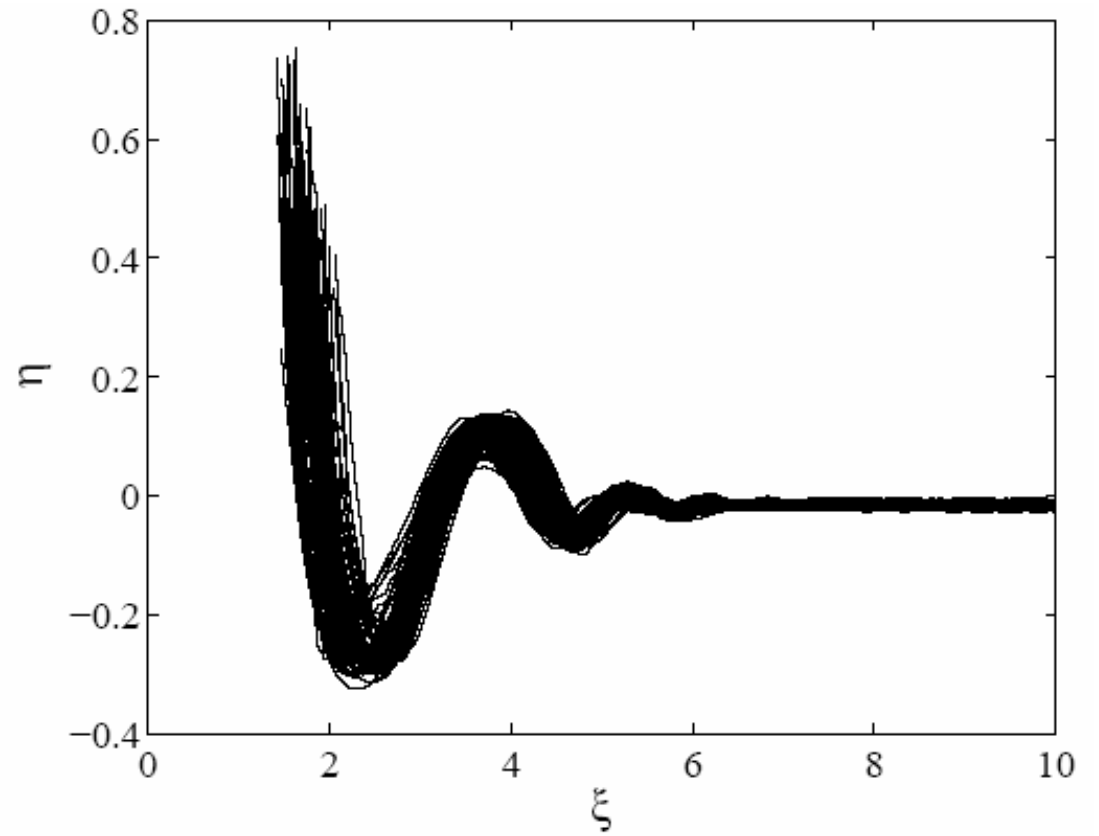
Extracting Wave amplitudes



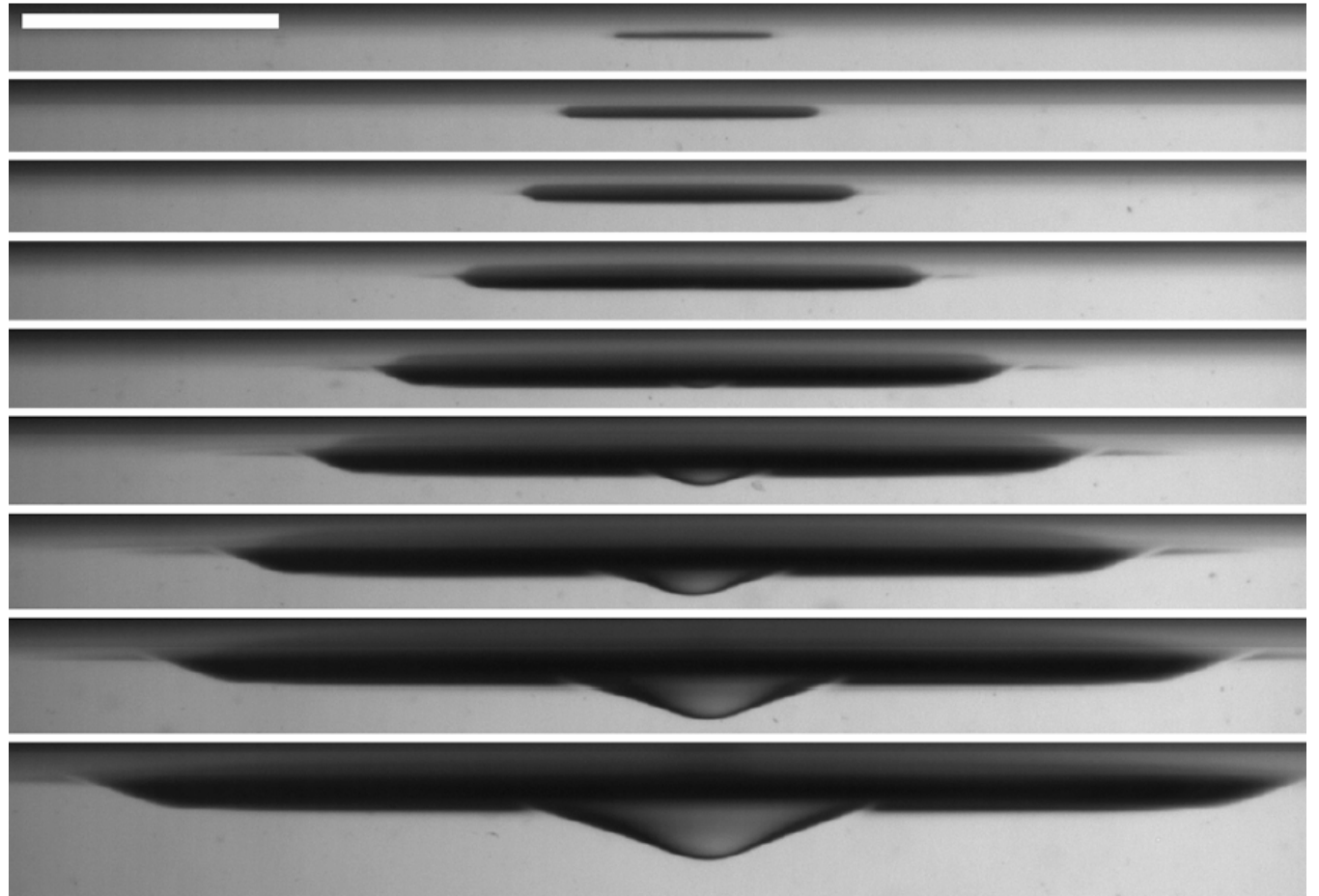
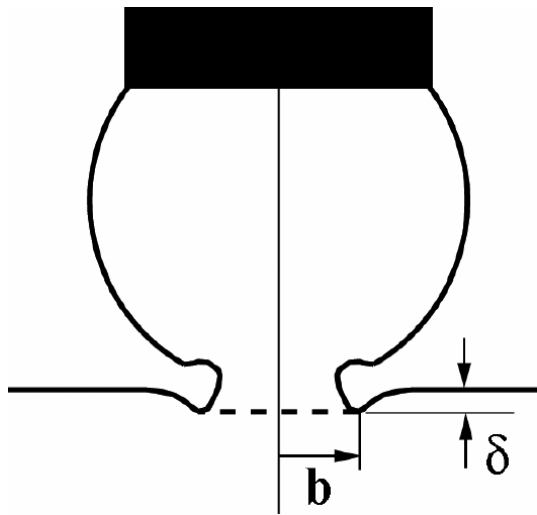




Self-similar Shapes



Keller & Miksis (1983)



Water cone

Entering ethanol pool

Overview and Conclusions

- Ultra-high-speed imaging required for small drops/bubbles
- Time-resolved imaging of pinch-off and coalescence
- Interesting capillary-inertial dynamics below to 1 μm length scale
- Possible dynamics below optical limit?